

## **40<sup>th</sup> IGS Lecture**

# ***Novel Techniques to Simulate and Monitor Contaminant- Geomaterial Interactions***

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Dr. D. N. Singh was born in 1965 at Shahjahanpur, UP, India. He has been a faculty member of Civil Engineering at Indian Institute of Technology Bombay, since 1994. Before joining IIT Bombay he has served Central Road Research Institute, New Delhi, and Indian Institute of Technology Kharagpur, for short durations. His early education was completed at Lucknow, UP, India. He obtained his bachelor's, masters and Doctoral degrees from IIT Kanpur in 1986, 1989 and 1993, respectively.

His teaching, and research and developmental activities are in quite diversified areas of **geotechnical engineering** (viz., soil mechanics, foundation engineering, environmental geotechnology, mechanics of unsaturated soils, soil characterization based on thermal and electrical properties, contaminant transport in porous media, mineralogical characterization, utilization and recycling of industrial waste, geotechnical centrifuge modeling, etc.). He has published **285** technical articles of which **199** are in the refereed journals. He has supervised **36 Doctoral (and 7 ongoing)** dissertations and **35 Masters** theses. He has been successful in filing **19 (Indian)** and **1 US patents** and **3** have been granted.

Apart from teaching and research, Dr. Singh has been very actively associated with some of the most prestigious business houses, as an in-house instructor and retainer consultant. He has taken up sponsored projects from Bhabha Atomic Research Centre (BARC), Indian Council of Agricultural Research (ICAR), Department of Science and Technology (DST-TIFAC), All India Council of Technical Education (AICTE) and Atomic Energy Regulatory Board (AERB). With financial aids received from these and several other organizations, he has established a state-of-the-art **Environmental Geotechnology Laboratory** in the department.

Dr. Singh has founded *Environmental Geotechnics*, ICE Publishing, London, UK, and has been its **Editor-in-chief**. He has been 'Editorial Board Member' of several journals of repute. He is recipient of *Young teachers' award* instituted by the AICTE, New Delhi and **JUNIOR/SENIOR Paper Award** from the **International Association for Computer Methods and Advances in Geomechanics (IACMAG)** for the year 2005. He was **Chairman, 12<sup>th</sup> IACMAG, GOA, India, 1-6 October 2008**.

He is recipient of **Excellent Contributions Award 2008**, which is given by the IACMAG to individuals who have a record of significant contributions in research, academic activities and professional service in different regions of the globe. He is recipient of **John R. Booker Excellence Award-2011**, which is given by the IACMAG for *advancement of research, education and practice of Environmental Geotechnology and development of novel techniques to simulate contaminant transport in geomaterials, under laboratory and in-situ conditions*. He is **recipient of Richard Feynman Prize 2014**, for the best paper published by the **ICE journal**.

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1. **Indian National Academy of Engineering, New Delhi**
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**Abstract:**

Generation of huge amount of the toxic & hazardous wastes coming out of various industrial and domestic activities is becoming a major threat to the society. In the long-run, mainly because of non-scientific storage, disposal and closure, and due to the presence of undesirable concentration(s) of chemicals and radio-nuclides, elevated temperatures and microbial activity, these wastes (read *contaminants*) interact with geomaterials viz., soils, rock mass, ground-water. This interaction, termed as *contaminant-geomaterial* interaction, depending upon the severity of the contaminant(s) and interaction time, might alter overall characteristics of the geomaterials. Unfortunately, conventional laboratory and field instrumentation techniques are not well equipped to capture such interaction(s) and the mechanisms that prevail in the geomaterials. Hence, to achieve these objectives, evolving adequate and workable strategies, and modalities, that are nondestructive, noninvasive and economical is desirable. In this context, author's association with several industries resulted in development of innovative, cost-effective, yet efficient *techniques* that facilitates laboratory and/or in-situ simulation and monitoring of such interaction(s). Details of these techniques, the philosophy behind their creation and the way they can be employed for safeguarding geo-environment, from deterioration, are presented and discussed in this paper. Also, a brief discussion on some of the real-life situations where such techniques can be applied easily, by suitably modifying them, is presented for the benefit of the aspiring researchers and professionals.

**Keywords:** Geo-environment, toxic and hazardous waste, waste disposal, contaminant, geomaterial, interaction, instrumentation, monitoring, simulation.

## 1. Introduction

Various industries, considered as the backbone of a nation and responsible for making it *powerful* and *self-reliant* viz., manufacturing, pharmaceutical, oil & gas exploration, thermal power plants, atomic power generation, research & medication installations, dredging and mining, generate huge amount of hazardous and/or toxic waste (Singh, 2014; Rao et al., 2014; Bo, 2015; Hoyos et al., 2015; Rajesh et al., 2015; Rowe, 2014, 2015). Furthermore, activities like agriculture, construction & demolition, and 'domestic discharges' are responsible for generation tremendous amount of the municipal solid waste, MSW. Keeping in view the *sustainable growth*, ideally, strategies to utilize such waste, as a *man-made resource*, should have been practiced (Arulrajah, 2015; Tripathy, 2015). In this context, it is heartening to note that these issues are being recognized by the researchers, planners, industries and the administration as a *major threat* for the society (Jayanthi and Singh, 2016; Koshy and Singh, 2016a,b; Jayanthi et al. 2017; Ganaraj and Singh, 2017; Shetty and Singh, 2017, 2018), and *ready-to-adopt techniques* that have potential to overcome this situation, are being developed, marketed and implemented.

However, in order to safeguard the ecosystem against the wrath of such a situation, the waste, which might inherently contain leachable heavy metals and radionuclides, designated as *contaminants*, should be isolated from the *geoenvironment* by proper containment, encapsulation, immobilization, vitrification and/or disposal in deep geological formations (IAEA, 2004). An easy way to visualize the *geoenvironment* is to think of the subsurface domain, the *geomicrobiological sphere*, that mainly comprises the soil and rock mass, microorganisms and ground-water (Reddy, 2015; Pathak et al., 2014a,b, 2016a,b; Johnston and O'Kelly, 2016; Rakesh et al., 2017). In the long-run, these *contaminants*, due to their *interaction* with the rain-water and/or fluctuating ground-water table, which act as a nemesis for the units (read canisters in case of the radionuclides) in which the waste is contained, tend to contaminate the *geoenvironment* (Marti, 2015; Benson, 2016).

Incidentally, attributes like chemical activity, high temperatures and the presence of radionuclides, in contaminants, and inherent characteristics of the geomaterials viz., mineralogy and the presence of organic matter, pose a great challenge in evolving a proper and workable strategy to deter geoenvironmental degradation. To add to this list of woes, which makes this situation more *complex*, but *realistic*, the microbial (bacterial, fungal, pathogenic, viral and algal) activities pose a major threat to the

geoenvironment. Incidentally, the author; a *geotechnical engineer* who as a student was averse to *biology*, was induced to take up this challenge by *bio-savvy* and *enthusiastic* young researchers (Sharma and Singh, 2015; Sharma et al., 2016; Shashank et al., 2016,2018; Sowmya et al., 2017; Meenu et al., 2017). Another challenge being faced, in this context, is the *multi-phase* state of geomaterials, which makes it *not-so-easy* task to *simulate* and *monitor* the influence of *contaminants* on *geomaterials* and the mechanisms prevailing in them (Dangayach et al., 2015; Joseph et al., 2016,2017). From this discussion, it can be realized that handling, storage and disposal of toxic and hazardous wastes, and identifying/establishing their impact on the geoenvironment (by conducting environmental impact assessment,EIA) is becoming an utmost priority (Gurumoorthy and Singh, 2004a,b; Gurumoorthy and Singh, 2005, Rakesh et al., 2009; Rao et al., 2008, Arnepalli et al., 2010, Hanumantha Rao et al., 2013).

For safe execution of geotechnical engineering projects, various laboratory and field *instrumentation techniques* and *testing protocols* that facilitate characterization of geomaterials have already been developed. And this knowledge has also been adopted and prescribed by various statutory agencies such as FHWA (Federal Highway Administration, USA), ASTM (American Society for Testing and Materials) International, USA, BS (British Standards), EC (Eurocode), BIS (Bureau of Indian Standards) and EPA (Environmental Protection Agency), in the form of *codal-provisions*. These protocols are quite useful to capture *macro-characteristics* of geomaterials viz., strength, compressibility and hydraulic conductivity, which are mainly driven by *mechanical stresses* (better designated as *stimuli*), only. However, for a situation like *contaminant-geomaterial* interaction, picturized above, capturing *micro-mechanisms* viz., diffusion, dispersion, sorption&desorption, decay/decomposition, deposition, erosion and suffusion, exo- and endo- thermic reactions, heat and moisture migration, cracking, rheology and coupling of these mechanisms, becomes very crucial (Goreham and Lake, 2018). Needless to state, these mechanisms are driven by *environmental factors* viz., humidity, temperature, solar cycle, wind speed and rainfall intensity, and might evolve due to chemical, thermal, electromagnetic, radiation and microbial activities. These factors can be designated as *environmental stimuli* and their influence on overall characteristics (physical, chemical, mineralogical and biological) of the geomaterials and the mechanisms that prevail in them, should be investigated rigorously when *contaminant-geomaterial* interaction occurs (Kadali et al., 2013a,b, 2014, 2016). Needless to emphasize, all this falls beyond the realm of the *conventional geotechnical*

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methodologies and techniques and hence resorting to an interdisciplinary approach becomes inevitable (Kibsgaard, 2016).

Furthermore, the situation becomes dire when reputed and established equipment manufacturers shy away in extending their helping hand to the research fraternity, due to several reasons and constraints viz., lack of interest, inertia to adopt and practice new concepts, fear of not fulfilling supply-demand chain, lack of financial viability, etc. However, this turned out to be *ablessing in disguise* and prompted the author and his research group (designated as *THE ENVGEOs*) to deal with simulation and monitoring *contaminant-geomaterial* interaction. This incidentally, was the starting point of our journey in the realm of geoenvironmental engineering (Singh, 2016). A brief account of the research & development contributions made by *THE ENVGEOs*, in context of simulating and monitoring short- and long-term *contaminant-geomaterial* interaction and *the prevailing mechanisms*, is presented in the following. In due course of time, these techniques, which are a trendsetter, indigenous, economical and ingenious, became a panacea for solving various geoenvironmental issues highlighting the fact that *there is nothing called a Problem, it's just absence of an idea to find, the Solution*.

## 2. Types of Contaminant-Geomaterial Interaction

Based on the above discussion, it becomes imperative that efforts should be made to simulate, and quantify, *contaminant-geomaterial* interaction, which would be very useful for addressing various *geoenvironmental issues* that the contemporary society is facing, in a precise manner. The changes undergone by the geomaterials, when contaminants (exhibiting attributes like elevated temperatures, chemical and radionuclide concentrations) interact with them. This would be quite important and crucial for designing structures such as liners of waste containment facilities, landfills, cores of the dam, which are primarily constructed from fine-grained soils, treatment and stabilization of highly contaminated soils and dredged sediments, stabilization of soft & sensitive soils, and construction of roads & airfields by resorting to thermal treatment (Rowe, 2015; Zhang et al., 2011; Shetty and Singh, 2017).

The possible situations when *contaminant-geomaterial* interaction are: (1) passive geomaterial interacting with water, which corresponds to minimal interaction viz., sand-water interaction, (2) active-geomaterial interacting with water viz., fine-grained soil-water interaction, which could result in heat of wetting, *HOW*, (3) passive

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geomaterial interacting with contaminants viz., sands-contaminant interaction, and (4) active geomaterial interacting with contaminants, viz., fine-grained soil-contaminant interaction. These interactions might occur, in *geoenvironment* either due to *natural or man-made activities*. Hence, the effect of these activities on overall properties of the geomaterials (pre- and post- interaction) should be investigated (Sreedeeep and Singh, 2005a, 2006a, 2011). In this context, though, researchers have employed various *techniques* and *instrumentation* viz., batch tests, column tests, lysimetric studies, geotechnical centrifuge modeling and impedance spectroscopy, to monitor and quantify *contaminant-geomaterial interaction*, under controlled environmental conditions, most of these studies are contaminant and geomaterial specific and hence cannot be generalized, especially when *soils* are exposed to higher concentrations and temperatures associated with the contaminants.

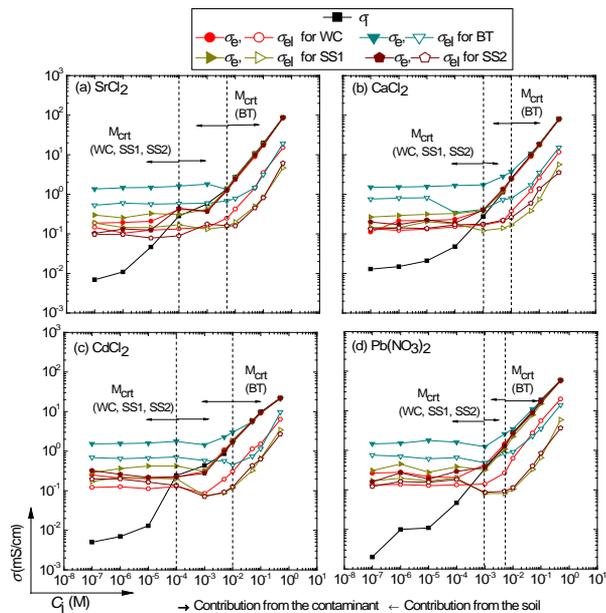
### 2.1 Quantification by using $k_d$ Parameter

The *contaminant-geomaterial* can be quantified by employing a parameter, designated as *distribution coefficient*,  $k_d$ . It has been demonstrated by Arnepalli et al. (2010), Pathak et al. (2014a,b) that several parameters viz., specific surface area, *SSA*, cation exchange capacity, *CEC*, percentage organic matter, percentage of fines, *CL*, mineralogy, redox potential, type of ionic species and their concentration, *pH*, temperature, liquid to solid ratio, *L/S*, and interaction time, influence determination of  $k_d$ , which requires very intricate experimental investigations. Hence, design of the *batch tests* to minimize the experimental efforts, and to understand the influence of so many parameters on  $k_d$ , becomes most desirable. Also, due to interplay of these parameters, it becomes difficult to identify the parameter(s) which would have most significant influence on  $k_d$ . It should be noted that identification of such parameter(s) would be helpful in optimizing the number of tests to be conducted for determining  $k_d$  of a *contaminant-soil system*, *CSS*, which otherwise is a very cumbersome task. To achieve this, *Taguchi method*, which primarily helps in designing experiments (Taguchi and Konishi, 1987) was employed for (i) identification of the most significant parameters on which  $k_d$  depends and (ii) *estimating*  $k_d$  by inputting these parameters, in the form of a *lookup table*, which would be quite handy for all those who design and analyze different types of *CSSs*, as described in the following.

Various soils such as White clay (designated as WC), Bentonite (designated as BT), five natural soils namely S1, S2, S3, SS1, and SS2, collected from different parts of

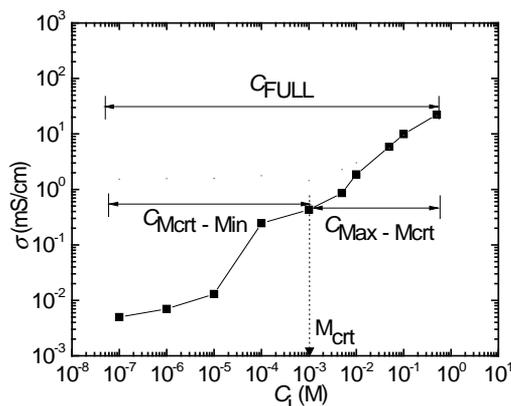
India, were used. Further, SrCl<sub>2</sub>, CaCl<sub>2</sub>, CdCl<sub>2</sub>, HgCl<sub>2</sub>, ZnCl<sub>2</sub>, FeCl<sub>3</sub>, NiCl<sub>2</sub>, and nitrates of Thorium, Uranium, Lead, and Cobalt in their non-radioactive form, were employed as the model contaminants. Batch tests were conducted following the guidelines presented in literature (ASTM D4646-03, 2008; Arnepalli et al., 2010) and various *sorption/desorption isotherms* were employed to compute the respective  $k_d$  (Pathak et al., 2014a,b). The effect of concentration of contaminants on  $k_d$  was standardized by employing *electrical conductivity*,  $\sigma$ , of the some of the *contaminant-soil solutions* during sorption and desorption processes, as depicted in Fig. 1. The data depicted by solid and open symbols, in the figure, correspond to sorption and desorption processes, respectively, for various CSSs.

It can be observed from Fig. 1 that, as expected,  $\sigma$  increases with an increase in initial concentration,  $C_i$ , of the contaminants used in the study. This highlights the fact that the contaminant contributes more in the value of  $\sigma$  of the CSS as compared to the soil.



**Fig. 1. Variation of electrical conductivity after sorption and desorption for different CSSs.**

However, the reverse phenomenon takes over when  $\sigma$  of the CSS increases due to leaching of the cations present in the soil, when it interacts with the lower concentration of the contaminant. These facts can be elaborated further by employing a term *critical concentration* or *critical molarity*,  $M_{crt}$ , depicted as a vertical dotted lines in Fig. 1. Incidentally, this line demarcates contribution of soils towards *desorption* (marked as a left side arrow, when soil is dominating in the CSS) and *sorption* mechanisms (marked as a right side arrow, when contaminant is dominating in the CSS). Hence, it can be opined that the concentration of the contaminant plays a very crucial and decisive role in determination of  $k_d$  of the CSS. As such, establishment of  $M_{crt}$ , by conducting batch tests with a wide range of concentration of contaminants and determining the sorption mechanism prevailing in a certain concentration range of contaminant, becomes mandatory to obtain meaningful  $k_d$ . To establish the influence of  $C_i$  on  $k_d$ , three cases of the  $\sigma$  versus  $C_i$  relationship (refer Fig. 2), which also exhibit the  $M_{crt}$ , were considered. These cases are: (i)  $C_{FULL}$ , (ii)  $C_{M_{crt}-Min}$  and (iii)  $C_{Max-M_{crt}}$ , as depicted in Fig. 2, where  $C_{FULL}$ ,  $C_{M_{crt}-Min}$  and  $C_{Max-M_{crt}}$  exhibit concentration ranges from  $0.5$  to  $1 \times 10^{-7}$  M,  $M_{crt}$  to  $1 \times 10^{-7}$  M and  $0.5$  M to  $M_{crt}$ , respectively.



**Fig. 2. The three cases of the initial concentrations of the contaminant considered for determination of the  $k_d$**

Furthermore, *kinetic and thermodynamic* studies for various CSSs were conducted on soils BT, WC, S1, S2, S3 and contaminants  $SrCl_2$ ,  $Pb(NO_3)_2$  and  $UO_2(NO_3)_2$ , which are the major isotopes present in radioactive waste, by performing batch tests. The kinetic reaction depends upon the interaction between the sorbate-sorbent (i.e., the contaminant-soil) system (Azizian, 2004), which also helps in determining the changes in the concentration of the sorbate over a certain duration. As such, the

sorption rate of the contaminants onto the soil particles (i.e., the sorption mechanism) was also studied. As temperature is an important factor that influences the sorption characteristics of the soil (Wang, 2011), batch tests were performed by varying temperature,  $T$  (viz.,  $27\pm 2^\circ\text{C}$ ,  $35\pm 2^\circ\text{C}$  and  $45\pm 2^\circ\text{C}$ ) while, the interaction time of CSS,  $L/S$ , soil  $pH$ , soil and contaminant species were maintained unchanged. The thermodynamic studies were performed to understand (i) how energy changes and/or dissipates in the CSS and (ii) the nature of reaction (i.e. spontaneous or non-spontaneous) within this system.

As mentioned earlier, *Taguchi method* was used to design batch tests by employing L25 orthogonal array, *OA*, referred by Taguchi and Konishi (1987) and Wu and Zhou (2011), which is a special matrix having optimal setting of maximum six parameters and minimum two parameters with five setting levels (i.e., the experimental conditions). It is based on the selection of the five parameters in the study (refer Table 1).

**Table 1. Selection of the input parameters and experimental conditions**

S.N.	$L/S$	$C_i$ (M)	Contaminant	CEC (meq/100g)	$pH_i$
1	10	0.1	SrCl <sub>2</sub>	105.05	1
2	20	0.02	CaCl <sub>2</sub>	82.41	3
3	50	0.003	CdCl <sub>2</sub>	77.39	5
4	100	0.0005	Pb(NO <sub>3</sub> ) <sub>2</sub>	54.58	6
5	200	0.0001	HgCl <sub>2</sub>	37.59	7

However, it should be realized that for investigating the influence of these five parameters on  $k_d$ ,  $5^5 (=3125)$  experiments need to be conducted, *which turns out to be a mammoth task*. This necessitates establishment of sensitivity of the  $k_d$  on the various attributes of the soil-contaminant system, as discussed above. Such a sensitivity analysis would be helpful in minimizing the number of experiments, otherwise required, to determine  $k_d$ . Moreover, for investigating the influence of these parameters on  $k_d$ , a wide range of experimental conditions were chosen for quantifying the *contaminant-soil* interaction. Subsequently, batch tests were designed for five different parameters and five experimental conditions, as mentioned in Table 1. Accordingly, L25 *OA* was found suitable, and only 25 batch tests were performed

for understanding the mutual influence of the five parameters. Based on this analysis, a *lookup table* (refer Table 2) for various CSSs was also developed. The *lookup table* is a quick interface for estimating  $k_d$  for different types of *contaminant-soil systems*, and can be utilized for proper design and execution of various geoenvironmental projects mentioned earlier.

The efficiency of the *lookup table* should be verified for several *contaminant-soil systems* by conducting batch tests. The methodologies based on *electrical* and *microwave* impedances should be developed for determination of  $k_d$ , a panacea for detecting the influence of the waste on the geoenvironment.

**Table 2. The “Lookup Table”**

Soil	pH	Soil characteristics							Contaminant characteristics				Output
		CL	SSA	CEC <sub>T</sub>	RWL	Minerals	MO	ξ	Cationic species	pH	T	T <sub>e</sub>	
BT		90	621	105	33.80	M	4.27	33.50	U, Th, Pb, Cd Ca, Sr Co, Zn Fe, Ni Hg	3 4 5 6 7 8	27 35 45 55 65	6 12 24 36 48 72	$k_d$ (l/kg)
WC	7.5	54	49	38	48.63	K	3.62	17.50					
S1	6.2	31	23	82	56.55	Ma, Mu	10.60	14.50					
S2	5.2	70	21	64	70.18	Mu, Bi MI	1.60	24.89					
S3	7.1	44	289	61	83.48	Mu	2.94	12.50					
SS1	7.3	0	629	77	42.73	Q, He Ma	4.55	19.21					
SS2	7.8	0	279	55	29.78	Q, C	7.66	27.51					

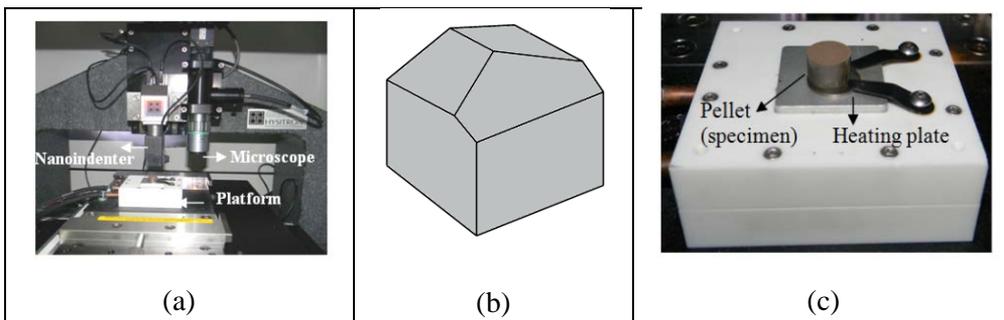
*Note:* CL: clay fraction (in %); SSA: specific surface area (in m<sup>2</sup>/g); CEC<sub>T</sub>: Total cation exchange capacity (in meq./100g); ξ: Zeta potential (in -mV); RWL: Reactivity with lime (in %); MO: Summation of CaO, K<sub>2</sub>O and Na<sub>2</sub>O (in %); T: temperature (in °C); T<sub>e</sub>: Interaction time (in hr.); M: Montmorillonite; K: Kaolinite; Ma: Magnetite; Mu: Muscovite; Bi: Biotite, MI: Mullite; Q: Quartz, He: Hematite; C: Calcite

## 2.2 Quantification by using XRD, Nano-indentation and LSD Techniques

Kadali et al. (2013a) attempted to: (a) develop a methodology, which can be employed for quantifying *contaminant-soil interaction*, under laboratory conditions, and (b) investigate the changes undergone by the soil, when it interacts with the contaminants (possessing both high temperature and concentration). To achieve this, the state-of-

the-art *instrumentation* and *techniques* viz., laser scanning diffraction, X-ray diffraction stress analysis and nano-indentation have been employed. Such a study has been found to be very handy to develop a *soil characterization scheme*, based on the parameters that imbibe physical, chemical and mineralogical characteristics of the soil, and can easily and quickly be determined in any conventional geotechnical engineering laboratory, as described in the following.

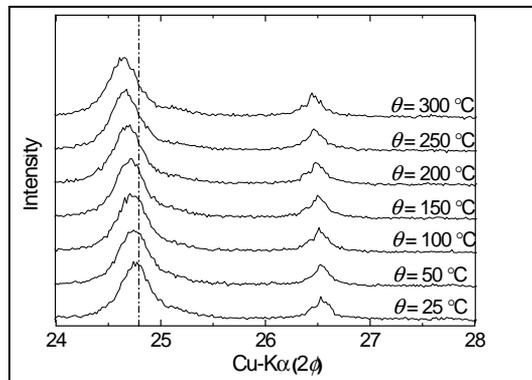
The mechanical and engineering characteristics (such as hardness, residual modulus, and resistance to indentation) of the geomaterial were determined up to 200 °C, in steps of 50 °C, by employing a nano-indenter (Model: TI 900 TriboIndenter, marketed by Hysitron, USA). This equipment contains a CCD camera, attached to an optical microscope (refer Fig. 3a). A Berkovich indenter, refer Fig. 3b, which is a three-sided pyramid of *radius of curvature* ~150 nm and measures the force imposed on the pellet (read specimen, refer Fig. 3c), was employed for these investigations. The corresponding displacements are recorded with the help of LVDTs.



**Fig. 3. The setup depicting (a) nano-indenter along with the microscope, (b) the Berkovich indenter and (c) the specimen mounted on the heating plate**

The investigations conducted by Kadali et al. (2013a) have revealed that the *changes* in the color of soils, heated up to 250 °C, can be attributed to depletion of the organic matter, primarily (Ulery and Graham, 1993; Parlak, 2011). However, corresponding to the temperatures greater than 250 °C, except for the soils containing kaolinite, this phenomenon could be attributed to an increased oxidation and other *chemico-mineralogical* changes (Kadali et al., 2013b). An increase in the *specific gravity* and decrease in the *specific surface area*, SSA, of the soil, due to its exposure to elevated temperatures, has also been observed. This phenomenon can be attributed to the loss of moisture, presence of the organic matter, impurities and changes occurring at

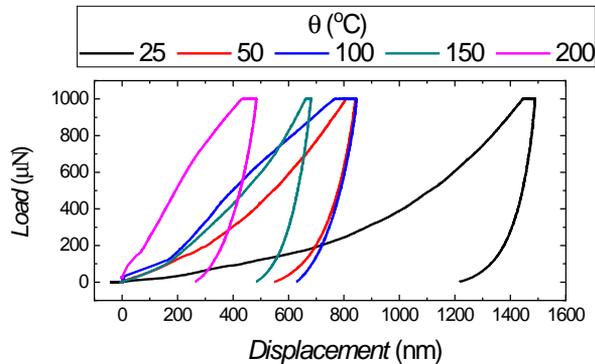
elemental level. It has also been observed that, except for the soils with passive minerals viz., quartz and kaolinite, the clay-sized fraction decreases while the silt-sized fraction increases with an increase in temperature. This observation further substantiates that exposure of the soil to elevated temperatures results in an increase in its particle size. The decrease in the SSA of the soil can further be substantiated by an increase in its particle size due to its exposure to elevated temperatures, as substantiated by *laser scanning diffraction* analysis. Interestingly, it has been observed that the cation-exchange capacity, *CEC*, of the soil decreases as the exposure temperature increases. This can be attributed to a reduction in the exchangeable cations and loss of organic matter present in the soil (Cox et al., 2000; Simkovic et al., 2008). Furthermore, a reduction in zeta potential (West and Stewart, 1995; Vane and Zhang, 1997; Yukselenand Kaya, 2003; Kayaand Yukselen, 2005; Akbulut and Arasan, 2010; Moayedi et al., 2011) of the soil, which defines its capacity to interact with the environment (read water, air, gases, contaminants), due to its exposure to elevated temperatures and excess contaminant concentrations, has also been observed (Chorom and Rengasamy; 1996; Ketterings et al., 2000; Abu-Zreig et al., 2001; Hatten et al., 2005; Liu, 2010; Yilmaz, 2011). The investigations by employing X-ray diffraction were instrumental in demonstrating, successfully, that with an increase in exposure temperature,  $\theta$ , the lattice spacing changes (refer Fig. 4), which indicates mineralogical *phase transformation*.



**Fig. 4. The X-ray diffraction patterns of the soil exposed to different temperatures**

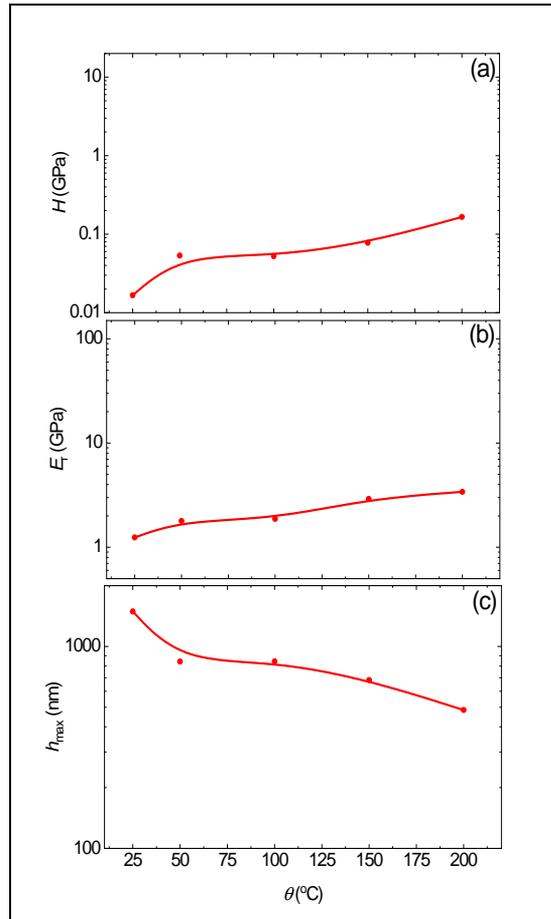
Here it is worth reiterating that the changes in crystallographic characteristics of the geomaterial would strongly influence its physical and chemical properties. It has also been observed that with an increase in temperature, the soils containing passive

minerals viz., quartz and kaolinite, exhibit *expansion*. On the contrary, soils containing active minerals viz., montmorillonite, *shrink*, due to the loss of *hygroscopic moisture* (Shah and Singh, 2006; Prakash et al., 2016). Furthermore, the investigations conducted by Kadali et al. (2014) were instrumental in demonstrating the utility of X-ray Diffraction analysis for establishing *residual normal* and *shear stresses* on the soil grains when they get exposed to the elevated temperatures (Tanaka et al., 1997; Sultan et al., 2002; Tan et al., 2004; Welzel et al., 2005). These researchers were also successful in demonstrating the potential of *nano-indentation*, which is mostly employed by material scientists for characterization of metals, in identifying the changes undergone by the fine- and coarse-grained soils when they get exposed to elevated temperatures (Penumadu et al., 2009; Daphalapurkar et al., 2011). A typical load versus displacement response of a soil exposed to different temperatures is depicted in Fig. 5.



**Fig. 5. Typical load vs. displacement response, obtained from the nano-indentation, for the soil exposed to elevated temperatures**

Based on our studies, we could also demonstrate that the fine-grained soils, in contrast to the coarse-grained soils, are more susceptible to changes in *hardness*,  $H$ , *residual modulus*,  $E_r$ , and *resistance to indentation*,  $h_{max}$ , when exposed to elevated temperatures (refer Fig. 6). However, efforts should be made to extend these studies by considering (a) the interaction of different types of contaminants and geomaterials of entirely different species, exposed to elevated temperatures, (b) quantification of such interaction and derivation of generalized relationships and (c) investigations related to the alteration in various geotechnical properties of the soils (viz., shear strength, compressibility, compaction characteristics, permeability etc.) due to such interaction(s).



**Fig. 6. Variation of the (a) hardness, (b) residual modulus and (c) maximum depth of indentation with respect to temperature for a typical soil**

### 2.3 Quantification by using Heat of Wetting

As mentioned earlier, the heat of wetting, *HOW*, is an interesting philosophy to quantify *contaminant (in fluid phase) and geomaterial* interaction. In case of the fluid that is a *liquid*, the setup employed for quantifying the *HOW* would be similar to the one used by professionals that work on *cementitious materials & admixtures* i.e., a calorimeter or a heat of hydration setup. However, when the liquid gets replaced with the *gas*, more intricate setups that would facilitate its *adsorption and/or sorption* on the

geomaterials (this process is a precursor to *sequestration*) need to be created or employed (Rajesh et al., 2015).

The *HOW* setup, to simulate *contaminant-geomaterial* interaction when the latter is in liquid form, consists of a calorimeter, which is a wide-mouth vacuum flask of 500 ml capacity, fitted at the top with a cork stopper containing two holes. Through one of these holes, a funnel for pouring the sample in the flask is fitted while the second hole facilitates fitting of a T-type thermocouple, which records temperature of the solution, when connected to a datalogger. The calibration of the thermocouple was carried out by inserting it in boiling water and ice. A stirrer, which is rotated by a motor at a constant speed, has been provided to facilitate proper mixing of the *soil* and *contaminant* in the liquid form. This simple setup, has been found to be quite useful for quantifying *contaminant-geomaterial* interaction, as explained in the following.

As depicted in Fig. 7, the temperature,  $\theta$ , versus time,  $t$ , trends for the soils of entirely different chemical constituents, could be utilized for computing the *HOW* of a geomaterial in its contaminated or uncontaminated states.

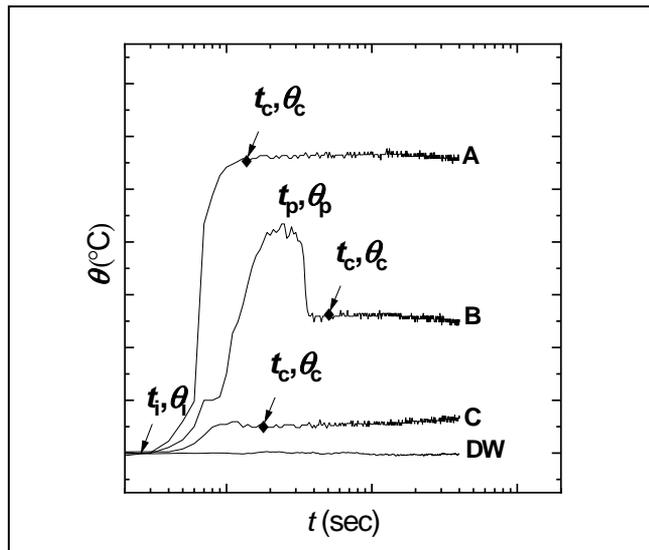
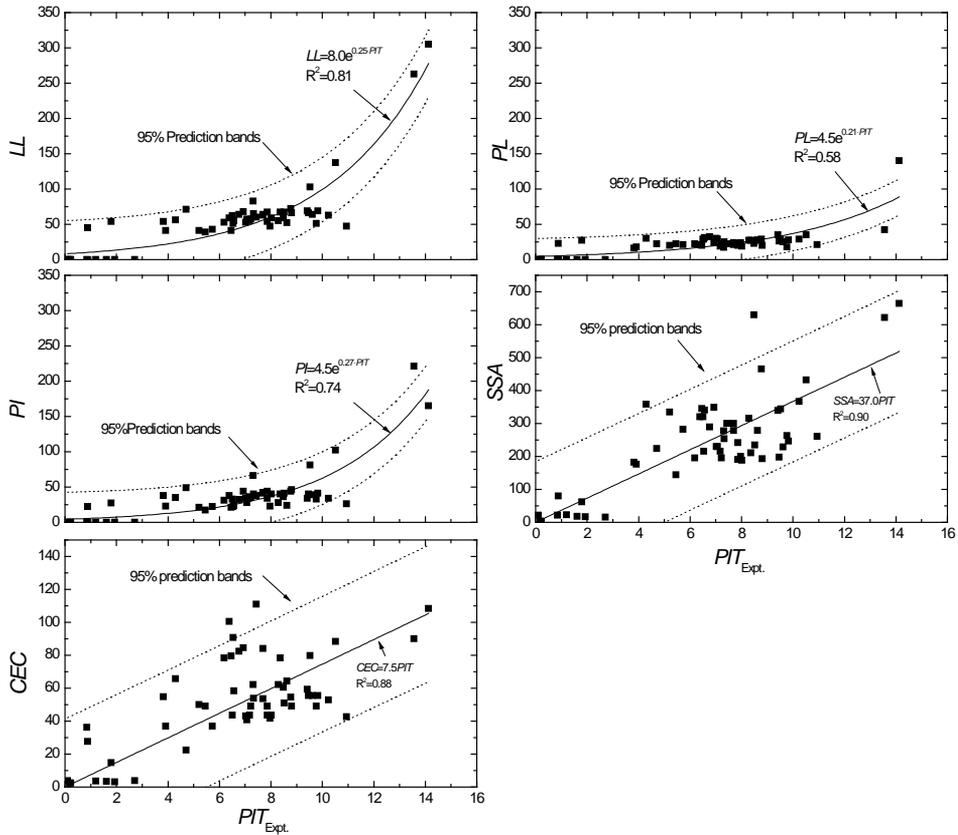


Fig. 7. The variation of temperature with time for *HOW* tests

As depicted in Fig. 7, the  $\theta$  versus  $t$  trends for the soils can be grouped into three categories: **A**, **B** and **C**, mainly depending upon their characteristics. While, both trends **A** and **B** exhibit a sharp increase in  $\theta$ , in case of the former, temperature attains almost a constant value,  $\theta_c$ , with time while in case of the latter, it drops down sharply from a peak temperature,  $\theta_p$ , followed by attainment of the constant temperature value,  $\theta_c$ . In author's opinion, these trends, in general, exhibit *soil-water interaction*, which primarily depends upon the physical, chemical and mineralogical characteristics of the soil (Bathija et al., 2009). However, trend **C** indicates very less interaction, as the changes in values of  $\theta$  are almost negligible. Subsequently, the percentage increase in the temperature of the soil-water mixture, *PIT*  $[(\theta_c - \theta_i) \cdot 100 / \theta_i]$ , can be computed. It should be noted that  $\theta_i$  corresponds to the ambient temperature, for  $t=0$  sec. Another parameter, the reaction time, *RT* ( $=t_c - t_i$ ), has been defined, which quantifies the minimum time required by the *soil-water* slurry to attain the constant temperature,  $\theta_c$ . It is author's hunch that *RT* is related to the *PIT* and hence it should be a measure of the potential of the soil to release heat (due to exo-thermic reaction) when it comes in contact with water. This study demonstrates the utility of the parameter, *HOW*, which has been quantified as the percent change in temperature of the fine-grained soils when they are allowed to interact with water (or for that matter any contaminant in the solution form) in a calorimeter. However, as during this interaction, the physical, chemical and mineralogical properties of soils play an important role, this parameter has been related to various soil specific properties viz., clay fraction present in the soil, *CL*, plasticity index, *PI*, liquid limit, *LL*, specific surface area, *SSA*, and cation-exchange capacity, *CEC*. Based on extensive investigations, on fifty soils of entirely different characteristics, it has been demonstrated that there exists a well-defined relationship between these parameters and the *PIT*, refer Fig. 8 (Kadali et al., 2016).



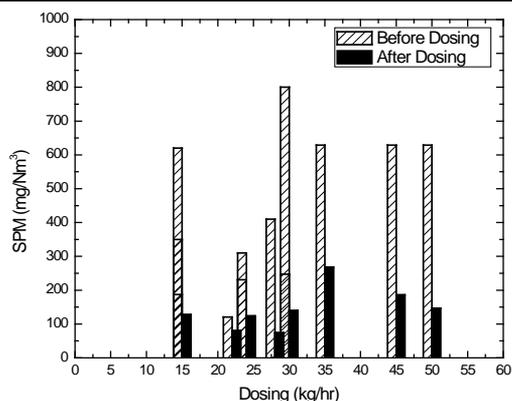
**Fig. 8. The variation of  $LL$ ,  $PL$ ,  $PI$ ,  $SSA$  and  $CEC$  with the  $PIT$  obtained from *HOW* experiments**

It is believed that the relationships proposed in Fig. 8 can be employed for determining the soil specific parameters, particularly the fine-grained soils, just by determining the  $PIT$ , which can be obtained easily, and quickly, by employing a calorimeter. Needless to mention, this philosophy (read technique) will be a boon for the researchers and professionals dealing with the characterization of soils. However, it is recommended that extensive studies should be conducted, by the research fraternity, to check the validity and utility of the proposed relationships (Fig. 8), and to refine them if they are found misleading. A big leap in this context would be to evolve a novel soil classification system for characterizing the soils by utilizing the parameter  $PIT$ , which includes in it the physical, chemical and mineralogical characteristics in a holistic manner.

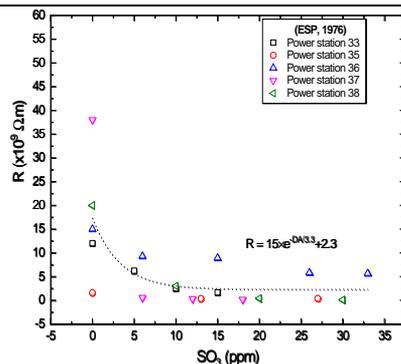
### 3. Gas-Geomaterial Interaction

Some good examples of *gas-geomaterial interaction* are: remediation of contaminated soils by air-sparging (Reddy and Adams, 2001), carbon sequestration (Sarhosis et al., 2018), radioactive waste disposal (Mahjoub and Rouabhi, 2018) and gas permeation in porous media (Rouf et al., 2016). Incidentally, determination of specific surface area, SSA, of geomaterials, by employing: (a) Blaine's air permeability apparatus (ASTM C 204) and (b) Absorption/Adsorption techniques, wherein Methylene Blue, Nitrogen gas, Ethylene Glycol Monoethyl Ether, EGME and Air, are employed (Arnepalli et al., 2008), is another good examples of *gas-geomaterial* interaction.

Another challenging but industry-relevant problems that Shanthakumar et al. (2008a,b) studied is *flue-gas conditioning*, FGC, of the fly ash, as described in the following. It is a well-understood fact that among various pollution control devices, electrostatic precipitators, ESPs, and cyclone separators, are popularly employed by the thermal power stations, and cement plants, for reduction of suspended particulate matter, SPM, as depicted in Fig. 9(a). This is mainly due to their greater efficiency in removing particles ( $<0.01 \mu\text{m}$  in size), their effectiveness in a wide range of operating temperatures, and their suitability for corrosive environmental conditions. However, the constraints associated with *implementing* various measures to enhance performance of ESPs at a power station are: (i) constraints associated with the feed coal (i.e., cost associated with import, washing of the coal and environmental issues associated with it, and ash content etc.), (ii) addition of more collection plate area in the ESP, which requires more space and is highly expensive, (iii) the installation and operating costs for employing wet ESPs is too high apart from lump formation of the ash, and its degradation as a construction material, (Singh and Kolay, 2002a). Under these circumstances, the FGC becomes the inevitable choice, mainly due to the fact that the (i) lower cost input compared to the establishment of additional ESPs, (ii) shorter execution time, (iii) more flexibility and versatility even if variations in operating parameters occur (such as coal variation, boiler load, ESP voltage and current change), SPM levels can be controlled/maintained easily by simply adjusting the amount of FGC agents. These agents are quite useful in reducing resistivity,  $R$ , refer Fig. 9(b) and hence improving the surface conduction characteristics of the particles of fly ash/dust (Shanthakumar et al, 2010).



**Fig. 9(a). Efficiency of the FGC at various power stations in India**

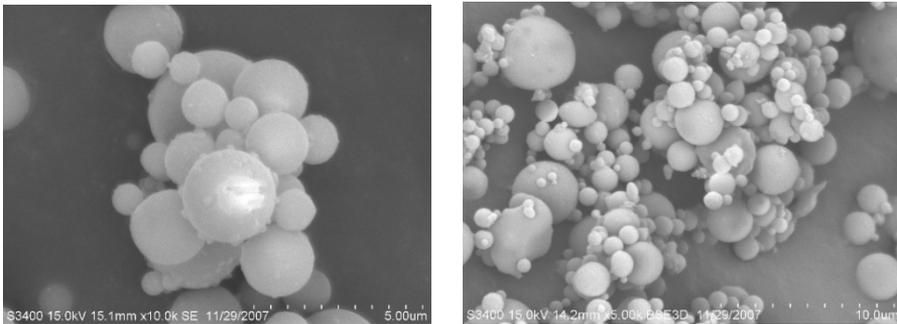


**Fig. 9(b). The effect of SO<sub>3</sub> dosing on resistivity of the fly ash**

Apart from this, researchers have demonstrated and critically evaluated the importance of the FGC by resorting to different techniques viz., sulphur trioxide conditioning, ammonia conditioning and dual flue gas conditioning, DFGC (Shanthakumar et al., 2009). It has been reported by these researchers that during conditioning of the flue gas, a part of the conditioning agent (particularly ammonia) gets *precipitated* on the fly ash particles, which might influence its overall characteristics and hence its *utility & marketability*. The conditioned ash might also lead to contamination of both the ground- and surface- water, if disposed in a landfill.

To answer these issues, comprehensive studies were conducted to investigate the influence of FGC on physical, chemical, mineralogical, morphological and pozzolanic characteristics of the fly ash. Also, how FGC results in improved performance of the ESPs, and hence in a reduction of the SPM emission levels, should be studied extensively keeping in view the intricate *gas-geomaterial interaction*, which is essential to design and model various environmental clean-up strategies.

In this context, the SEM micrographs of the fly ash (refer Fig. 10), after DFGC, reveal that this treatment is more effective for increasing the fly ash collection due to agglomeration of the ash particles, which enhances the collection efficiency of the ESP. This in turn results in less SPM emissions from the power plants levels (Shanthakumar et al., 2008a,b,2009,2011).



**Fig. 10. Micrographs of the fly ash after undergoing FGC**

However, the growing emphasis on utilization of the fly ash, and cenospheres, has raised major concerns about the suitability of the conditioned fly ash in cement and concrete industry (Bartake and Singh, 2005,2007; Das et al., 2011). This is mainly due to the fact that the presence of residual conditioning agent might adversely affect the characteristics of the fly ash. Hence, it becomes essential to ascertain the extent of changes undergone by the fly ash in its physical, chemical, mineralogical, morphological and pozzolanic characteristics, as a result of the FGC. Also, the influence of FGC on the *collection efficiency* of the ESPs needs to be investigated, which would be of great help in developing a suitable strategy for controlling the SPM emission levels at thermal power stations.

In this context, studies conducted by Shanthakumar et al. (2008a,b,2009,2011) turned out to be a big relief, for the power plants. These studies were instrumental in demonstrating that the *optimal dose* of the conditioning agent(s) does not alter the *lime reactivity* and *pozzolanic activity* (Dalinaidu et al., 2007) of the conditioned fly ash.

#### **4. Effects of Contaminant-Geomaterial Interaction**

When *contaminant-geomaterial interaction* occurs, depending upon the severity of this interaction, the geomaterial might undergo: (a) mineralogical alteration and (b) decomposition. These *micro-mechanisms* might result in *fabric* alteration of the geomaterial, as well, as explained in the following.

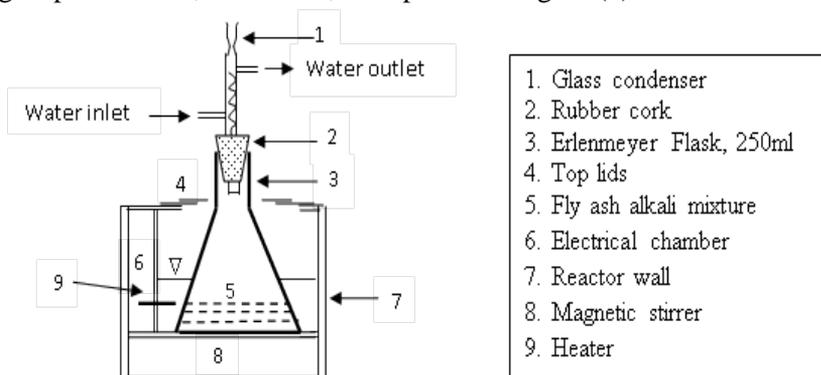
#### 4.1 Mineralogical alteration

One of the good examples of the *contaminant-geomaterial interaction* is ‘alkali activation’ of the fly ash, which results in mineralogical alteration of the fly ash (Singh and Kolay, 2002a). Hence, apart from many applications of the fly ash, known to the construction industry (Kolay et al., 2001, Kolay and Singh, 2002a,b; Singh et al., 2002), its valorization in the form of ‘fly ash zeolites’, FAZ, has gained significant attention of researchers in the recent past (Kolay and Singh, 2000,2001a, Kolay et al., 2001; Jha and Singh, 2011).

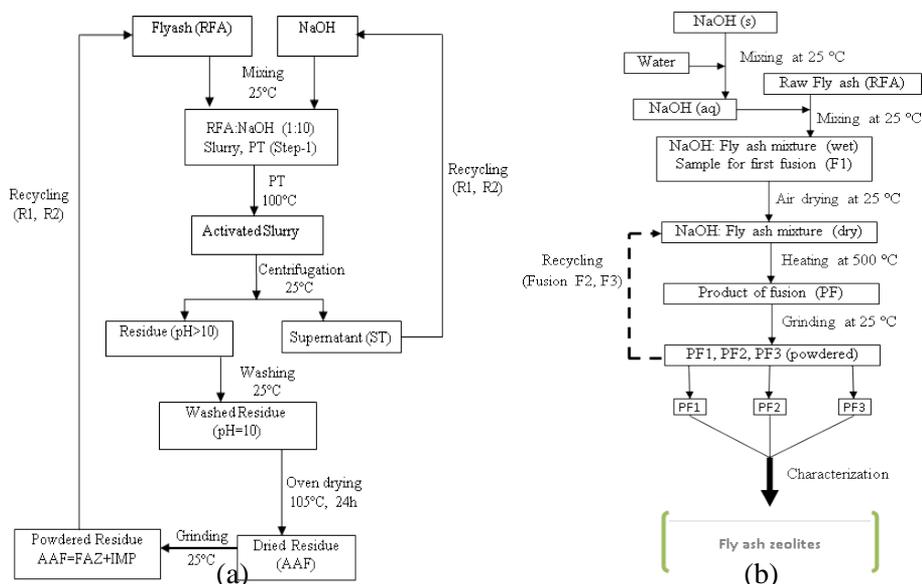
The FAZs viz., Na-P1, hydroxysodalite, faujasite, chabazite, analcime and cancrinite, are available in hydrated alumino-silicate mineral forms and can be synthesized from the fly ash by its *adequate*alkali activation. However, the physical properties viz., specific gravity, specific surface area, chemical composition viz., silica and alumina contents, and mineralogy viz., quartz and mullite, of the fly ash largely influence this process, the grades of FAZs formed, mostly quantified by the cation exchange capacity, in particular (Kolay and Singh, 2001b; Arnepalli et al., 2008; Bayat et al., 2015). Accordingly, use of fly ash zeolites in industries, especially for removal of heavy metal ions and other wastes for various environmental cleanup projects, would primarily depend upon their *grade*, which would mainly depend upon the method adopted for the synthesis (Kolay and Singh, 2002a,b; Koshy et al., 2015a,b; Shankara et al., 2016; Koshy and Singh, 2016a). Over the years, several methods: (a) hydrothermal, (b) fusion prior to hydro-thermaltreatment, (c) microwave assisted hydrothermal treatment and (d) molten salt technique have been adopted by researchers for achieving the desired *mineralogical alteration* of the fly ash, read synthesis of ash zeolites, (Jha and Singh, 2011,2012; Koshy and Singh, 2016b; Koshy et al., 2018), as discussed in the following.

To simulate the effect of *contaminant-geomaterial* interaction on *mineralogical alteration* of the geomaterial, the mixture of the raw (or original) fly ash, RFA, and different stock solutions of NaOH, was hydrothermally activated at 100 °C, in an open refluxed system, as depicted in Fig. 11. The activation duration was varied from 12 to 48 hours, with an increment of 12 hours (Jha and Singh, 2013,2014a,b). After each step of interaction, the residues and the supernatant were processed and recycled, as depicted in Fig. 12(a). The oven-dried residues from primary treatment, PT, were powdered, with the help of mortar and pestle, before each recycling step, R1 and R2.

Both the residues and supernatant from each step were characterized, simultaneously. On the other hand, the three-step activation, TSA, by fusion of the fly ash, coated with NaOH (NaOH/RFA ratio varied from 0.2 to 1.4, at an increment of 0.2), was conducted at 500 °C for 2 hours, in a Muffle furnace (Jha and Singh, 2014a). To overcome the issues associated with inadequate contact between the RFA and NaOH, fused residues were powdered using mortar and pestle, before subjecting them to recycling steps of fusion, F2 and F3, as depicted in Fig. 12(b).

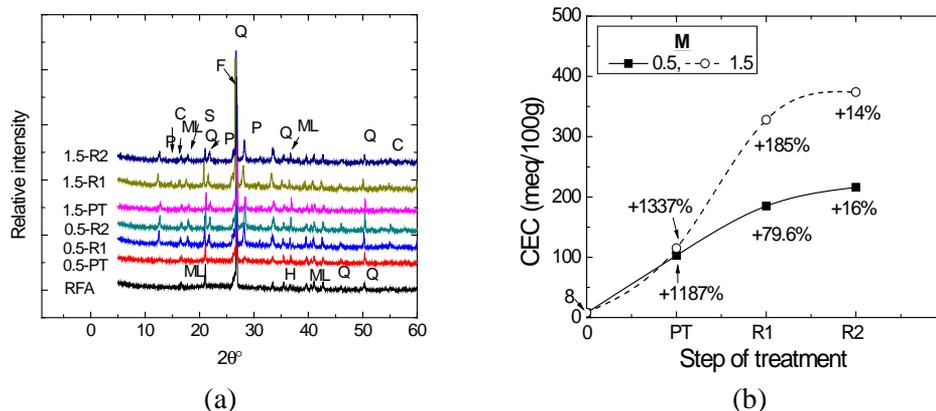


**Fig. 11. Details of the reactor employed for hydrothermal activation of the fly ash**

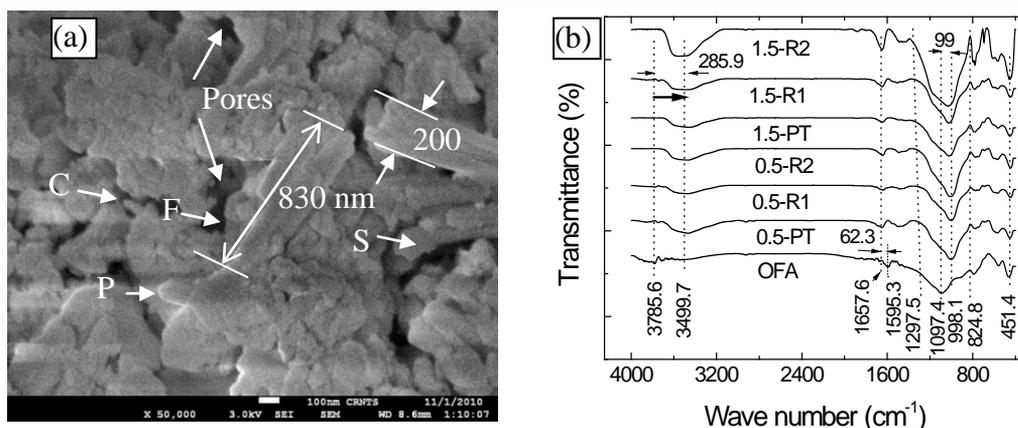


**Fig. 12. The stepwise flow charts for the TSA of the fly ash by (a) hydrothermal and (b) fusion methods**

The results obtained from these syntheses are presented in Figs.13 and 14. Symbols Q, ML, P, C, S, F, H and M designate the quartz, mullite, zeolite Na-P1, cancrinite, hydroxysodalite, faujasite, hematite and molarity of the NaOH solution, respectively. The designations 1.5-PT, 1.5-R1 and 1.5-R2 correspond to the residues produced after Step-1, 2 and 3 of the TSA, respectively, by using 1.5M NaOH and 12 hours of activation in each step. The formation of polycrystalline phases of the FAZs can be observed in Fig. 13a, which can be attributed to significant gain in the cation exchange capacity, CEC, of the residues (up to 185%) after the Step-2 (refer Fig. 13b). Although, the Step-3 causes less increase in the CEC, remarkable growth in the crystal shapes up to 830 nm (refer Fig. 14a) is indicative of enrichment with hydroxyl bonds and more -Si-O-Al- bonding (confirmed by a shift in the peak depicted in Fig. 14b). This indicates more crystallization in the residues, 1.5-R2 (refer Fig. 14b), as compared to the raw fly ash, RFA, (Eisazadehet al., 2012; Jha et al., 2015).



**Fig. 13. The effect of the TSA on characteristics of the residues, changes in the (a) mineralogy and (b) CEC**



**Fig. 14. The effect of the TSA on characteristics of the fly ash residues as confirmed from the (a) Morphology of 1.5-R2-24 and (b) FTIR results.**

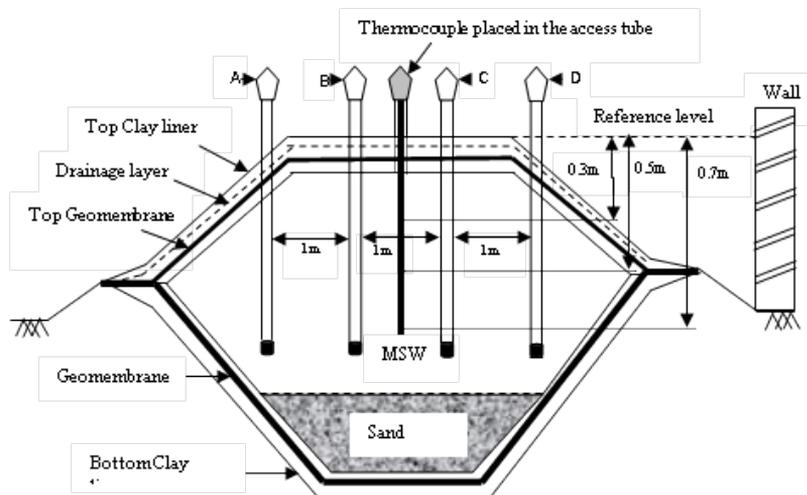
However, efforts should be made to relate the cation exchange capacity of the geomaterial with the silica-alumina ratio. Such a correlation would be very much useful for predicting the susceptibility of a geomaterial to get mineralogically altered. This philosophy would also be useful for the production of *synthetic minerals* at industrial scale for their application in various environmental clean-up projects.

## 4.2 Decomposition

The severity of the contaminants, in terms of elevated temperature, the concentration of the chemicals, radionuclides, microorganisms, and environmental radiations are primarily responsible for decomposition of geomaterials that consist of high organic matter viz., humus, peats, marine clays, biochar, municipal solid waste etc. This process might result in generation of leachates, semi-solids and gases, which tend to migrate through the appropriate *pathways* in the far- and near- fields to contaminate the geoenvironment (Rakesh et al., 2009,2017). The decomposed geomaterial should be treated as a ‘multi-phase system’ and its conceptualization and modeling is the prime focus of *The ENVGEOs*, presently, as discussed in the following.

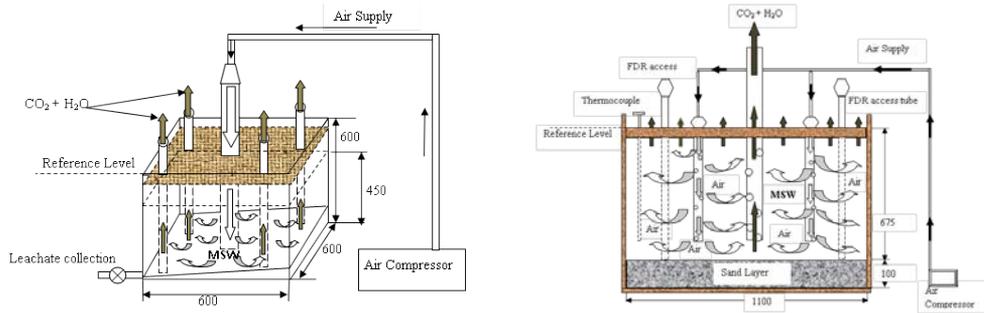
The major factors responsible for controlling the rate of decomposition of the municipal solid wastes, MSW, which author treats as the *manmaderesource*, are the moisture content, temperature and biological activities. To study the degradation characteristics of the MSW, sustainable engineered bioreactor landfill, SEBLF, were

developed. As depicted in Fig. 15, a scaled-down field model of the SEBLF would facilitate: (a) demonstration of the effectiveness of the state-of-the-art *instrumentation* viz., time and frequency domain reflectometry, TDR & FDR, probes and *multi-level* thermocouples, that is conceived for monitoring the moisture content and temperature, respectively, (b) to maintain the favorable conditions for bacterial activity, which would accelerate the decomposition rate by controlling the moisture content and temperature of the MSW inside the SEBLF, through scheduling of leachate recirculation, (c) to ascertain the controlled emission (and capturing) of biogas generated due to decomposition of MSW, (d) to analyze the leachate characteristics, which are indicative of the degree of decomposition, over a period of time, and (e) to assess groundwater and surrounding soil contamination (Patil and Singh, 2015; Patil et al., 2017; Chembukavu et al., 2019).



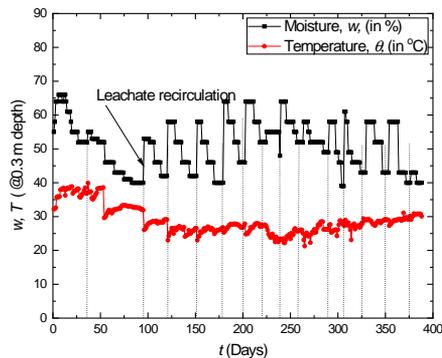
**Fig. 15. Installation of the FDR probe and thermocouple in the SEBLF**

To simulate decomposition of the MSW, under laboratory conditions, the (a) laboratory bioreactor landfill, designated as LBLF, (ref. Fig. 16a) and (b) the SEBLF, (ref. Fig. 16b) were developed. These setups have been found to be quite useful for conducting studies related to decomposition of the MSW.



**Fig. 16 (a) Schematic of the (a) laboratory bioreactor landfill, LBLF, and (b) sustainable engineered bioreactor landfill, SEBLF.**

The TDR & FDR probes, and thermocouples were employed for monitoring & controlling the moisture content,  $W$ , and temperature,  $\theta$ , at a certain depth in these reactors, respectively (refer Fig. 17). It should be noted that due to the spatial heterogeneity of the MSW, in its composition and density, the application of such probes and interpretation of the obtained results is a challenging task (SushaLekshmi et al., 2018a,b). Through this exercise, it has been realized that by proper scheduling of leachate recirculation (Reddy et al., 2016), the rate of decomposition of the MSW can be augmented up to a great extent (Patil et al., 2017). Furthermore, it has been demonstrated that due to the presence of various *macro- and micro- nutrients* in the *final harvested material* (the residues) of the anaerobic and aerobic BLFs could be an appropriate replacement for the conventional manures and fertilizers (Reddy et al., 2003; Patil et al., 2017).



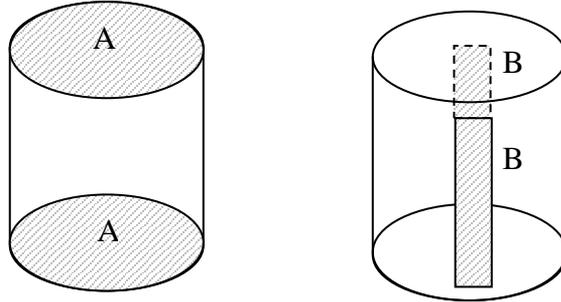
**Fig. 17. Monitoring decomposition of the MSW by employing state-of-the-art instrumentation and leachate recirculation**

However, extensive studies are required, to study the effect of heterogeneity of MSW on the decomposition process in the bioreactor landfill. Development and application of the low-cost wireless sensors in the bioreactor landfills, which can be used for monitoring the pH, moisture content, temperature, constituents of the biogas and bacterial growth appears to an idea worth trying. Also, *leachate augmentation* by treating it with microbes should also be tried.

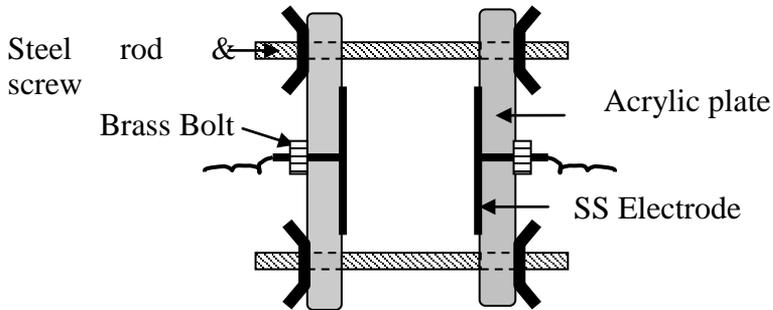
### 4.3 *Fabric Alteration*

Based on the discussion presented above, it is easy to comprehend that when *contaminant-geomaterial* interaction occurs, the *fabric* of the geomaterial might get changed (Fan et al., 2017). This is valid particularly for the particulate geomaterials viz., soils mass, which consists of the particles and pores of various sizes and shapes, and their arrangement, nature and distribution is termed as *fabric*. These parameters, and hence the *fabric*, influence engineering properties of the soil mass viz., shear strength, compressibility, rheology, conductivity of fluids, etc., to a great extent (Gumaste et al., 2014a,b; Du et al., 2015; Jha and Sivapullaiah, 2016). Hence, quantification of the *fabric*, and its anisotropy, becomes important for a better understanding of their overall response to *external stimuli*, particularly in case of the fine-grained soils (Gumaste et al., 2013; Du et al., 2014). In this context, a methodology based on electrical impedance measurement, termed as *Impedance Spectroscopy*, IS, for determining the *fabric* of the soil mass in its undisturbed, remolded, sedimented and *exposed to thermal flux* states was developed (Kaya et al., 2006; Rohini and Singh, 2004; Shah and Singh, 2004,2005,2006; Sreedeeep et al., 2004; Bhat et al., 2007; Gumaste et al., 2014a,b; SushaLekshmi et al., 2016). Incidentally, this methodology has also been found to be extremely useful for quantifying the *fabric*-anisotropy, as discussed in the following.

The undisturbed samples of the marine clays, were collected from the eastern coast of India, with their depth ranging from 5 to 65 m, below the seabed. To obtain *fabric* characteristics of these samples, their electrical conductivity,  $\sigma$ , was measured across **A-A** (longitudinal direction) and **B-B** (transverse direction) as depicted in Fig. 18. The test setup depicted in Fig. 19 was developed and employed for this purpose (Gumaste and Singh, 2010).

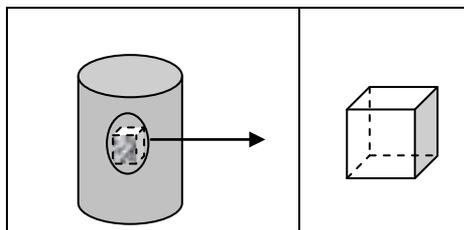


**Fig. 18. Measurement of the impedance response of the sample on (a) longitudinal (A-A), and (b) transverse (B-B) planes**



**Fig. 19. The sample holder used for conducting impedance spectroscopy of the sample**

Investigations to establish *fabric* of the sample were also conducted by resorting to scanning electron microscopy, SEM, and mercury intrusion porosimetry, MIP, on 5 mm cubical specimens, retrieved from the UDS or reconstituted samples, as depicted in Fig. 20.



**Fig. 20. Extrusion of the cubical specimen for establishing its *fabric***

These investigations reveal that *fabric* of the soil mass, and its anisotropy, could be quantified in terms of the electrical anisotropy viz., the anisotropy coefficient,  $A_e$ . As such, in case of the soil mass exhibiting *flocculated fabric* (i.e., the random orientation of the particles)  $A_e$  will be equal to (or approaches) unity. On the contrary, in case of the soil mass exhibiting *dispersed fabric*, the  $A_e$  is much higher than unity ( $>1.75$ ). In case of the undisturbed marine clays, normally,  $A_e$  is very high ( $\approx 2$ ) and it almost remains constant with the moisture content. However, in case of the samples retrieved from the same borehole, but from different depths,  $A_e$  is found to be strongly dependent on the sampling depth (the overburden). Incidentally, it is possible to classify or identify the type of *fabric* exhibited by the soil mass based on its  $A_e$  value, refer Fig. 21, which presents results obtained from the techniques like IS, SEM and MIP. Fig. 21 can be utilized as a guideline for determining the *fabric* (i.e., its type and corresponding mean pore diameter,  $d_m$ ) of the fine-grained clays, if  $A_e$  is known. Incidentally, it should be appreciated that determination of  $A_e$ , by adopting the technique mentioned above, is quite easy, and less time consuming, as compared to SEM and MIP techniques.

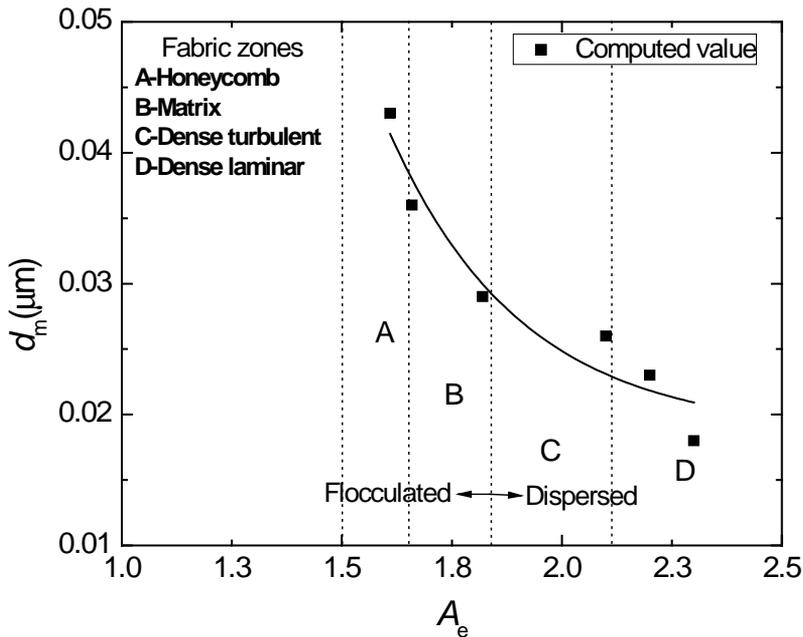


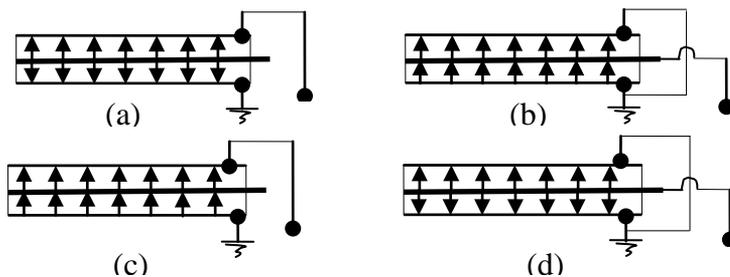
Fig. 21. Guidelines for determining *fabric* of the fine-grained soils

Hence, the advantage of IS over MIP and SEM, for determining *fabric* of the soil mass, in a nondestructive and noninvasive manner, and that too without facing the complexities associated with the sample preparation has been established. It must also be appreciated that the size of the sample and its moisture content are not a constraint for conducting the IS, as compared to its counterparts, and hence the results obtained would be more reliable.

#### 4.4 Alteration in Geomechanical Properties

It has been demonstrated by earlier researchers (Kulkarni et al., 2010; Patel et al., 2012) that various geomechanical properties of the soil mass viz., undrained shear strength, compression index, void ratio, elastic modulus and Poisson's ratio, can be determined in a noninvasive and nondestructive manner, easily, by correlating them to shear and compression wave velocities,  $V_s$  and  $V_p$ . Conventionally, resonant column, cyclic triaxial and torsional shear tests, which are quite intricate and expensive setups, have been employed for this purpose. However, *piezo-ceramics* (bender and extender elements) have been found to be very efficient in transmitting and receiving shear and compression waves (Bartake and Singh, 2007; Bartake et al., 2008) without impacting the sample, as strains produced are  $<0.001\%$ .

With this in view, a *Piezo-ceramics* were developed from the Lead Zirconate Titanate (LZT) based material, which exhibits high dielectric constant (Bhat et al., 2007), with high piezoelectric sensitivity, for low power consumption. Depending upon the requirements, these elements can be polarized in the same or opposite directions, as depicted in Fig. 22.



**Fig. 22. Different configurations of piezo-ceramics: *benders* (a) opposite-series and (b) same-parallel, and *extenders* (c) same-series and (d) opposite-parallel**

The block diagram depicted in Fig. 23 was developed for transmitting (T) and receiving (R) waves and proper functioning of the piezo-ceramics was confirmed by hearing a ‘sing’ sound when excited with a sine-wave pulse from the function generator(Bartake et al., 2008; Patel and Singh, 2009; Patel et al. 2009).

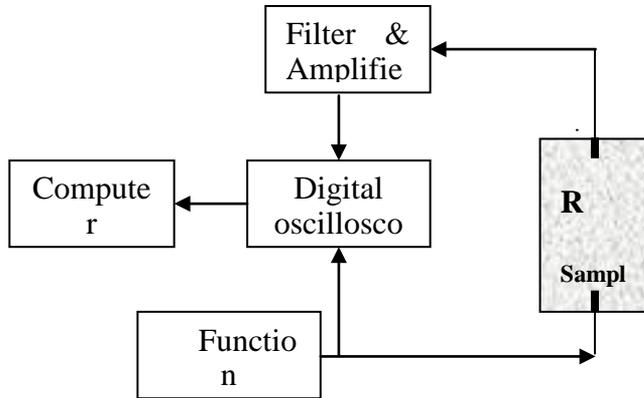


Fig. 23. The arrangement for transmitting and receiving waves in the sample

In order to check the versatility of these *piezo-ceramics*,  $V_s$  and  $V_p$  were measured on standard materials and also in asphaltic concrete and very soft clays (Patel et al., 2011).  $V_s$  and  $V_p$  were also measured for different types of soils in their remolded and undisturbed samples, UDS, by employing the setups depicted in Fig. 24.

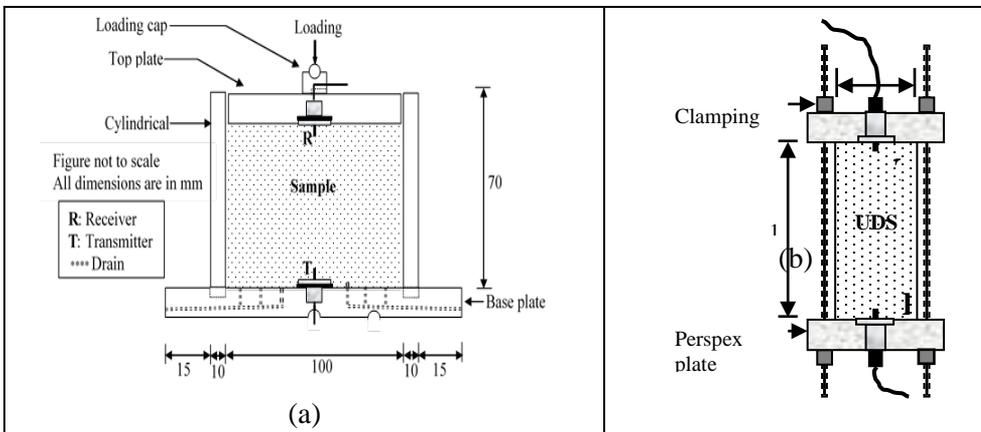


Fig. 24. The test setups used for (a) remolded and (b) undisturbed samples

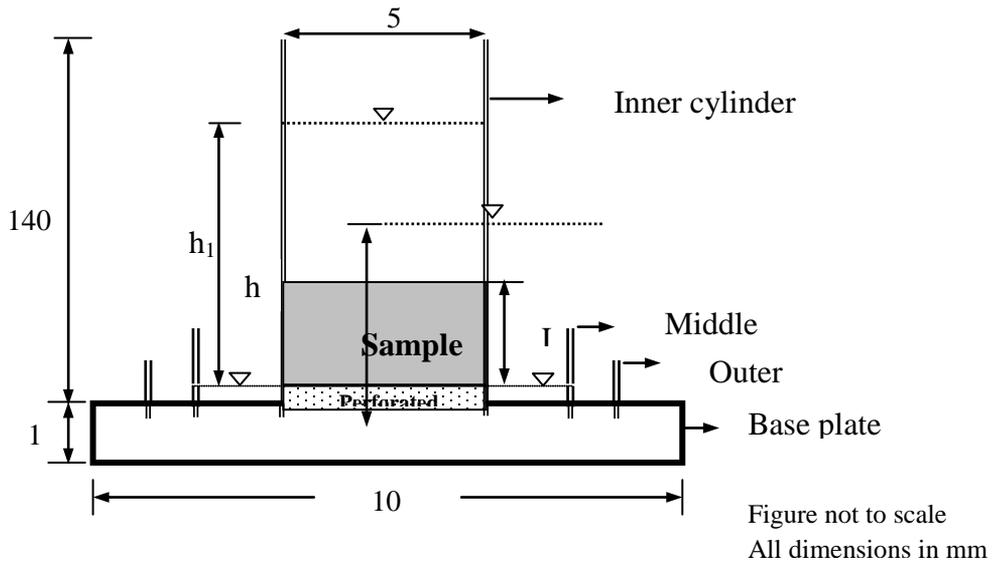
The above-mentioned discussion highlights that *piezo-ceramics* have already been utilized to characterize fine-grained and granular geomaterials. However, this technique should be employed to *detect* and *monitor* various micro-mechanisms viz., *mineralogical alteration*, *decomposition* and *fabric alteration*, that prevail in the geomaterials (in their saturated, unsaturated, contaminated and multi-phase) when they interact with contaminants. In this direction, it would also be quite prudent if electrical properties (resistivity and dielectric dispersion) of the geomaterials could also be linked with the  $V_s$  and  $V_p$ .

## 5. Monitoring and Simulation of Contaminant Transport

The *contaminant-geomaterial* interaction would also result in transport of contaminants in the soil and rock matrix. However, this interaction would be guided by, primarily, (a) the type and condition of the rock mass (intact or fractured) and (b) the soil mass (saturated or unsaturated). The fractured rock mass and the saturated soils create an easy pathway for the groundwater movement and thereby enhancing the probability of contamination of the geoenvironment (Gurumoorthy and Singh, 2004a,b; Singh et al. 2014,2015,2016). To counter this scenario, many a times, *immobilizing agents* (Arnepalli et al., 2010), tested for their *sorption and desorption characteristics* are employed (Pathak et al., 2014b,2016a,b; Moghal et al., 2017). Usually, *batch tests* are conducted to achieve this, but due to high liquid to solid ratios, these tests fail to simulate real-life situations. Also, these tests fail to come up with recommendations regarding selection of the most suitable and general isotherm, which would describe *sorption and desorption characteristics* of geomaterials, precisely (Arnepalli et al., 2010; Pathak et al., 2014a,2016a,b). This necessitates determination of the *sorption-desorption characteristics* of geomaterials by conducting 'column tests'. However, due to extremely low hydraulic conductivity of the geomaterials, column tests would require large durations to yield results and hence *geotechnical centrifuge modeling* has been conducted (Singh and Gupta, 2000,2001; Bartake et al., 2006).

The setup depicted in Fig. 25 was employed for determining hydraulic conductivity of the sample. This setup, when spun in a geotechnical centrifuge, simulates a *conventional falling-head permeability test* in an accelerated environment. The difference between the hydraulic conductivity tests and the column tests, which facilitate *contaminant-geomaterial* interaction, in a controlled manner, is that for

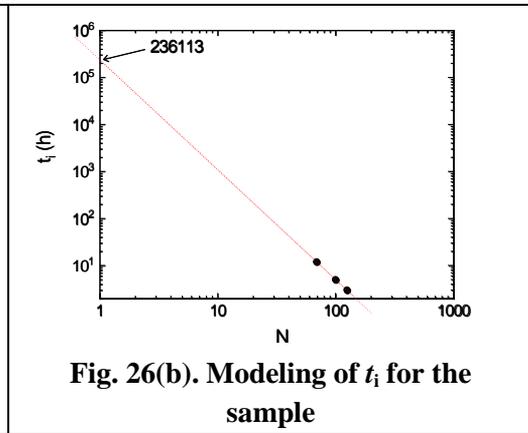
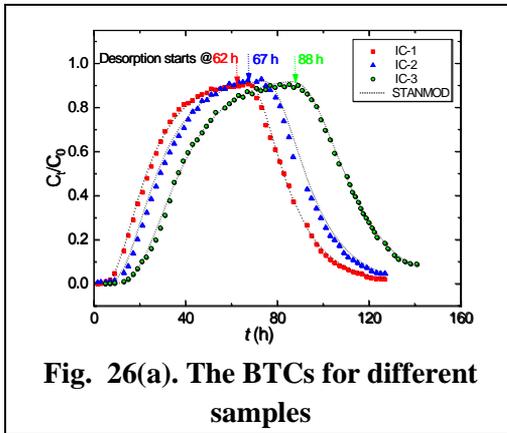
performing the later, the inner cylinder was filled with the *model contaminant*. The contaminants could be different ionic solutions in their active (radionuclides) and/or *inactive* forms. Hence, with the help of the setup depicted in Fig. 25, the *sorption* characteristics of the *contaminant* on *geomaterial* could be established. Subsequently, when the *contaminant* in the inner cylinder is replaced by the distilled water, the *desorption* characteristics of the *contaminant-geomaterial* interaction could be simulated and monitored.



**Fig. 25. The setup employed for conducting column tests**

Subsequently, the experimental results can be employed in STANMOD (<https://www.ars.usda.gov/pacific-west-area/riverside-ca/us-salinity-laboratory/docs/stanmod-model/>), to establish the break-through curves, BTCs, which represent the variation of normalized concentration of the contaminant in the fresh water with respect to time), which quantify *contaminant-geomaterial* interaction for a given contaminant and the geomaterial. In Fig. 26a, IC-1, IC-2 and IC-3, correspond to 8 mm, 10 mm and 14.5 mm thick Israeli chalk sample, centrifuged at 125-g, 100-g and 69-g, respectively. It can be noted from the Fig. 26a that for these samples,  $C_t/C_0$  (where  $C_0$  is the initial concentration of the contaminant and  $C_t$  corresponds to the concentration of the effluent after time,  $t$ ) attains almost unity after 62 h, 67 h and 88 h of centrifugation time, respectively. This time, denoted as  $t_{eq}$ , corresponds to the

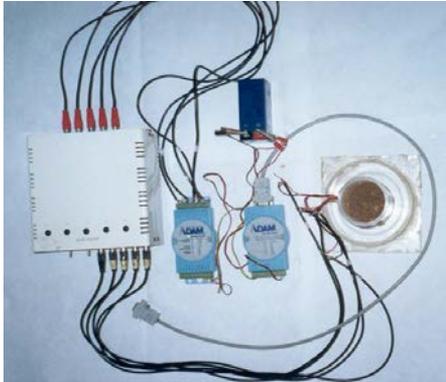
dynamic equilibrium sorption state of the material. Furthermore, it can be observed from Fig. 26(a), that initial BTCs get exhibited only after 3 h, 5 h, and 12 h of centrifugation for samples IC-1, IC-2 and IC-3, respectively. This indicates that the time,  $t_i$ , required for initial breakthrough to occur depends, apart from the retardation coefficient of the sample (material), on its length,  $L$ , and acceleration level,  $N$ . Further, in order to obtain  $t_i$  for 1 m thick rock mass, under normal conditions (i.e.,  $N=1$ ), the concept *modeling of models* was employed. The intercept of  $t_i$ - $N$  trend, depicted in Fig. 26(b), on the ordinate ( $N=1$ ) is found to yield  $t_i=236113$  h (i.e., 27 years). This demonstrates the utility of the geotechnical centrifuge (also known as accelerated physical modeling) modeling, for simulating and modeling *sorption-desorption* mechanisms, in an extremely short duration.



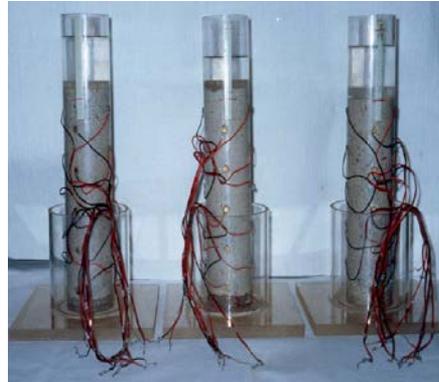
From this study, it has been demonstrated that the scale factors for the sorption and desorption times are 1.0 and 0.5, respectively. It is believed that the sorption and desorption parameters, obtained from the centrifuge modeling can be employed for evaluating the efficiency and suitability of immobilizing agents, and suitability of the geological formations (repositories), where the toxic/radioactive waste should be disposed.

In order to determine the presence and concentration of contaminants in geomaterials, a contaminant detector (refer Plate 1) has been developed by Rajeev and Singh (2004) and Kumar and Singh (2005). This detector helps in assessing the movement of contaminant front in a soil column (refer Plate 2). The movement of the *contaminant front* is done by sending a low-frequency sinusoidal signal into the sample, which is

connected to one arm of a Wheatstone's bridge. Hence, with the help of the contaminant detector, concentration of the contaminant at different locations (where electrode pairs are placed, diametrically opposite to each other) along the length of the sample can be detected (Kumar and Singh, 2005; Kumar and Singh, 2012).

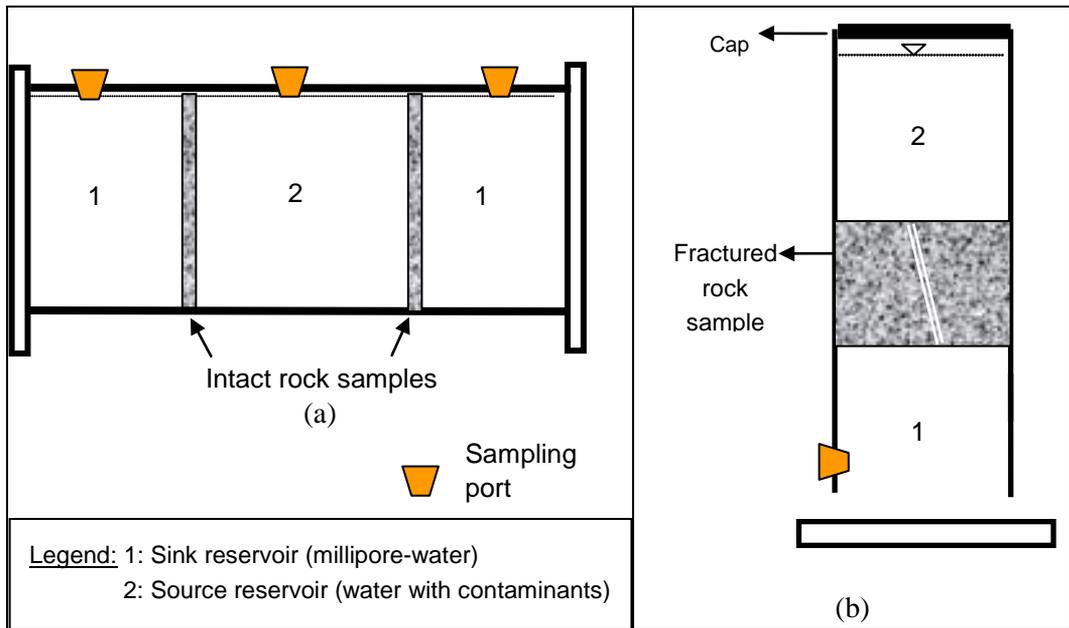


**Plate 1. The soil contaminant detector**

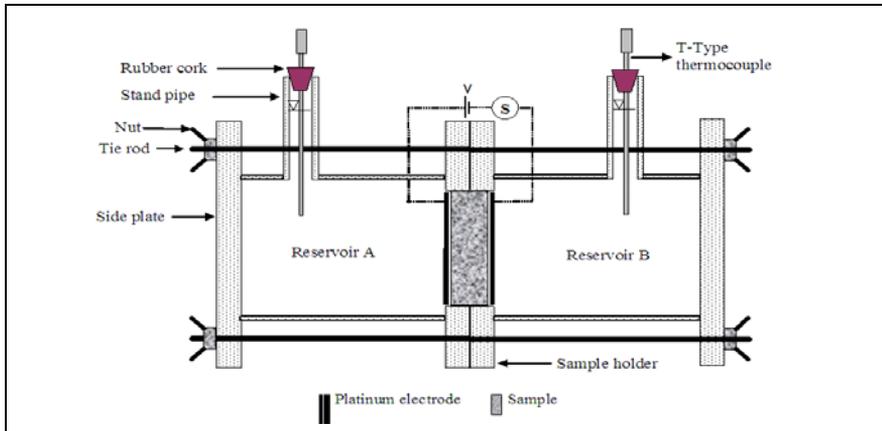


**Plate 2. The soil columns with contaminant placed at their top**

Furthermore, to study the mechanism(s) of contaminant transport in the rock mass viz., diffusion and/or advection, investigations were conducted by Gurumoorthy and Singh (2004a,b,2005) and Witthüser et al. (2006) by using different ions in their *active*(radionuclides) and *inactive* forms. Diffusion cells, depicted in Fig. 27(a)and (b) were developed for monitoring and establishing the response of the intact and fractured rock samples, respectively. Incidentally, Rao and Singh (2008) modified the diffusion cell depicted in Fig. 27(a) to accelerate the *contaminant-geomaterial* interaction that results in *diffusive contaminant transport*. This diffusion cell, refer Fig. 28, enhances the ionic diffusion through the rock sample by applying a suitable electric field across it.



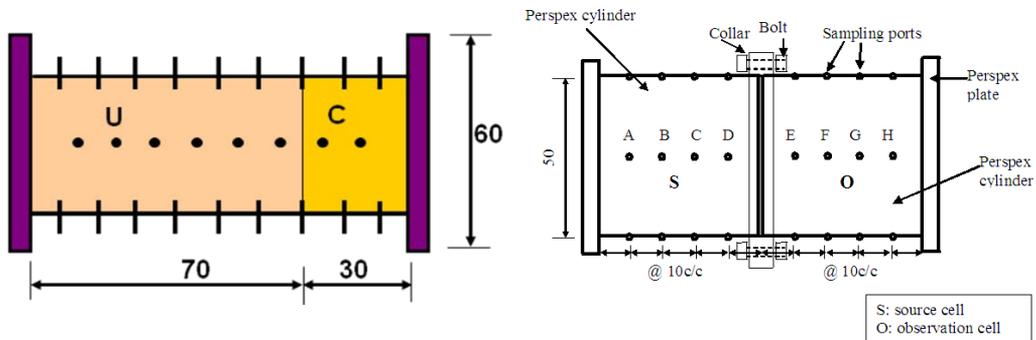
**Fig. 27. The diffusion cell for the (a) intact rock sample and (b) fractured rock sample**



**Fig. 28. The diffusion cell for accelerated tests on intact rock samples**

Furthermore, Sreedeeep and Singh (2008) fabricated a diffusion cell (refer Fig. 29, for investigating the *contaminant-geomaterial* interaction when diffusive contaminant transport occurs in unsaturated soils. The BTCs have been obtained by conducting the impedance spectroscopy, IS (Rao et al., 2007a,b). This non-destructive and non-invasive technique has been found to be an excellent way of studying contaminant transport in unsaturated soils.

However, for establishing diffusion characteristics of the saturated and unsaturated soil mass, Rakesh et al. (2009) have employed the diffusion cell, depicted in Fig. 30, with sampling ports placed at regular distance on the surface of the cells. These simple studies, which employ indigenously developed diffusion cells, are found to be quite handy for preparing guidelines for safe disposal of the radioactive waste in the geoenvironment.



**Fig. 29. The diffusion cell developed by Sreedeeep and Singh (2008)**

**Fig. 30. The diffusion cell developed by Rakesh et al. (2009)**

## 6. Monitoring and Simulation of Heat Migration

As discussed earlier, *contaminant-geomaterial* interaction might also result in generation of heat, mainly due to the chemical reactions. A good example of such a situation is the MSW in landfills, wherein several chemical reactions keep occurring. Furthermore, situations like waste from nuclear and thermal power plants, underground crude oil storage tank, oil-carrying pipelines, air conditioning ducts, buried conduits and electrical cables, geothermal energy piles, furnaces, boiler units, forging units, brick kilns, solar pond and rocket launching pads (Mondal et al., 2018a), necessitate understanding the influence of heat energy on geomaterials (Yun and

Santamarina, 2008). Moreover, remediation of the contaminated land and the thermal stabilization of the soil mass and sludge sediments, underground explosions and natural freeze-thaw cycle persistently subject the migration of thermal energy in geomaterials (the soil and rock mass). Incidentally, the activities like agriculture, wherein the yield can be accelerated by employing a set of parallel buried pipes as a heating system, also induce heating of the soil mass. In all these situations, it becomes mandatory to measure the rate and extent of heat migration in the soil mass. This can be obtained by establishing the thermal regime i.e., the temperature and heat flux profiles for a prolonged duration., with the help of *sensors* viz., single needle probe, dual probe heat pulse (DPHP) and flux sensors (Palaparthi et al., 2013,2017). The knowledge of thermal response of geomaterials also has a major role to play in safe and appropriate design of the underground and thermo-active structures (Abuel-Naga et al., 2015; Hytiris et al., 2017; Pellet, 2017).

In this context, efforts made by earlier researchers towards development of various techniques viz., flux plate method, calorimetric, gradient & combination, guarded hot plate, divided bar apparatus, cylindrical configuration, in-situ sphere, heat meter, thermal cell, thermal probe method (Mondal et al., 2018c), are worth appreciating. The MEMS (Micro-electro-mechanical system) based thermal sensor, differential scanning calorimeter, which are applied to sense the thermal response (say thermal properties) of the soil mass viz., thermal conductivity, resistivity, diffusivity, and volumetric heat capacity, are also being used. Though these methods and techniques are being used by the researchers and practitioners for thermal characterization of geomaterials to understand its heat migration capability, they are time-consuming, complex, tedious and costly. Also, Moreover, these methods and techniques do not imbibe the influence of microbial activities, organic content, mineralogical characteristics and heterogeneity of the geomaterials, and the MSW. It should be noted that most of these attributes of the geomaterials are sensitive to temperature and hence would have a definite influence over their thermal response, mostly at temperatures different than that prevailing at the ambience. This calls for extensive studies to establish the mechanism of heat migration in geomaterials, by measuring their thermal response i.e., thermal regime, particularly, under variable temperature conditions. One of the ways to achieve this would be to quantify ‘the zone of influence of the heat source’, by measuring the thermal flux,  $\phi$ , and temperature,  $\theta$ , over prolonged durations in the spatial domain (Mondal et al., 2016,2017,2018b). This could be done

quite efficiently with the help of flux sensors and MEMS-based temperature sensors (Susha Lekshmi et al., 2014), of extremely high precision.

As depicted in Fig. 31, the test setup for studying heat migration in soil mass consists of a mold, a hot air gun for inducing heat to the sample and a thermocol box. The hot air gun used in this study was manufactured by STEINEL thermo-power, Germany (model HL 2010 E) and it operates from 50 °C to 600 °C. The sample was prepared in a mold of dimension 150 mm×150 mm×200 mm. The four sides of the mold were made of 8 mm thick perspex and the base was made of a 10 mm thick aluminium plate. The top face of the mold (refer Fig. 31) was kept open to facilitate the flow of heat flux into the sample. After achieving a certain height of the sample, a flux sensors were embedded. It was ensured that the sensing face of these sensors faces the heat gun. A soil cover of 50 mm was ensured on the top most sensor for its proper functioning. Similarly, beneath the bottom-most flux sensor, a 30 mm thick layer of soil was maintained, which ensures 1-Dimensional heat flow in the system (refer section ABHG in Fig. 32). Subsequently, four thermocouples (T-type), were embedded in the sample to measure the temperature at the locations where heat flux sensors were installed. The response of the soil mass, corresponding to hot air gun temperature,  $\theta_{hg}$ , 80 °C is depicted in Fig. 33.

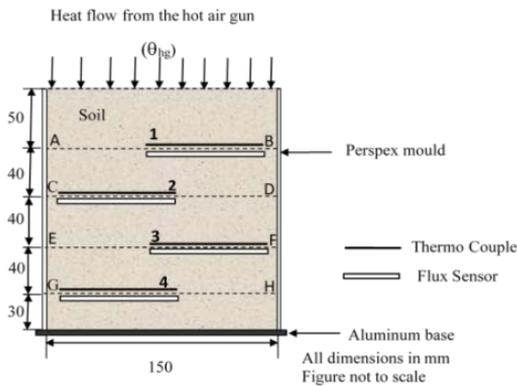


Fig. 31. Details of the test setup

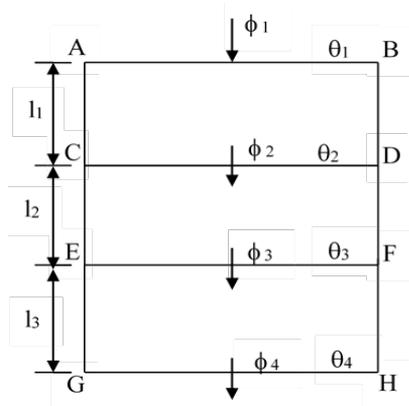
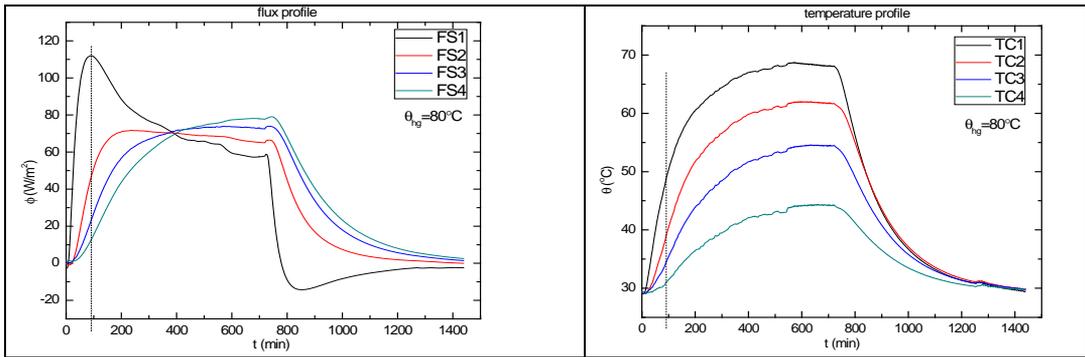


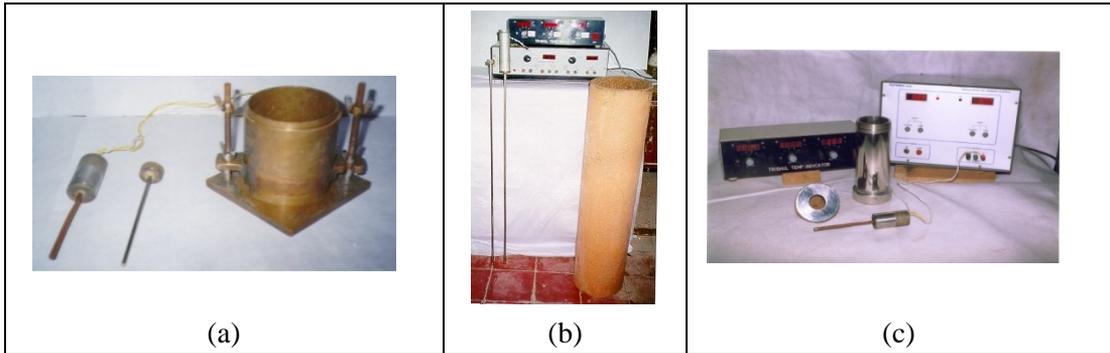
Fig. 32. Details of various sections of the sample



**Fig. 33. The variation of thermal flux and temperature, with time in the sample**

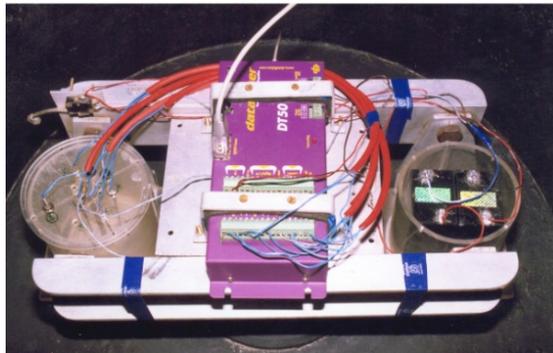
Apart from this, simple numerical approach, based on the finite difference method, has also been developed to solve the one-dimensional heat conduction equation, which facilitates establishment of the thermal regime in the soil mass. Also, a unified mathematical framework, ATHERES, has also been developed to estimate thermal regime in the dry soil mass. The proposed approach couples Taylor's series into the governing differential equation of one-dimensional heat conduction (Mondal et al., 2017).

As described earlier, *geomaterial-contaminant* interaction might result in endo- or exothermic reactions and/or a coupled flowviz., heat-induced moisture and contaminant migration (Poulose et al., 2000; Krishnaiah and Singh, 2003b; Bao et al., 2016). For conducting such experiments, thermal probes (refer Plate 3a,b) and a thermal property detector, TPD, (refer Plate 3c) were developed (Singh et al., 2003; Dalinaidu and Singh, 2004a,b). These probes and the TPD have been found to be very helpful in determining the thermal response viz., thermal resistivity, thermal diffusivity and effusivity, and specific heat, of different geomaterials, corresponding to different moisture contents and dry densities (Gangadhara Rao et al., 1998, 1999; Singh and Gangadhara Rao, 1998; Singh and Devid, 2000; Singh et al., 2001a; Dalinaidu and Singh, 2004a,b; Sreedeeep et al., 2005; Krishnaiah and Singh, 2003a,2004a,2006a). Krishnaiah et al. (2004) and Das et al. (2008a,b) have employed these instrument for determining the porosity of the rock-and concrete-samples, respectively, very successfully.



**Plate 3. The (a) laboratory and (b) field thermal probes and (c) thermal property detector**

Incidentally, heat migration in soil mass was simulated and monitored, in a geotechnical centrifuge, (refer Plate 4) by Manthana and Singh (2001) and Krishnaiah and Singh (2004b, 2006b), the validity of ‘modeling of models’ for heat migration in soils has been demonstrated, successfully. This study also demonstrated that the geotechnical centrifuge, can be used as a research tool to model heat migration in geomaterials.



**Plate 4. Instrumentation for heat migration in soils (centrifuge tests)**

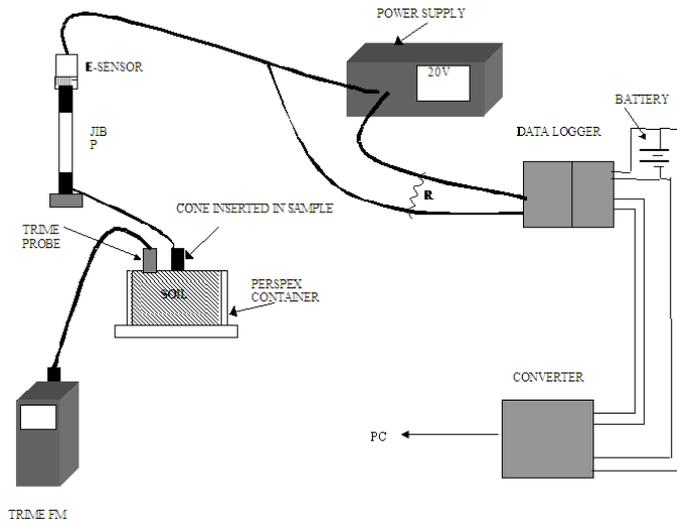
These studies are a good testimonial of the fact that cost-effective and indigenously made, simple looking devices viz., thermal probes and TPD, can be utilized for determining thermal properties of geomaterials such as thermal resistivity, thermal

diffusivity and specific heat, which are *must to determine* for proper design and safe execution of the projects mentioned above. Such in-house developments, also facilitate in demonstrating the facts; (a) thermal resistivity of geomaterials decreases as their dry-density and the moisture content increase, (b) the fine-grained soils exhibit higher thermal resistivity as compared to their counterparts, the coarse-grained soils, (c) thermal diffusivity of the geomaterials is found to be practically independent of their dry-density, (d) thermal diffusivity of the coarse-grained soils, which is maximum corresponding to the optimum moisture content, is higher than the fine-grained soils, for which the influence of the molding moisture content is practically negligible.

However, these developments should be employed, and if required augmented, for their applications to investigate (a) thermal response of multi-phase geomaterials viz., frozen geomaterials, gas hydrates and MSW, and (b) coupled phenomenon, wherein the moisture migration occurs due to the imposed thermal gradients.

## 7. Role of Environmental Stresses: Genesis of Unsaturated Geomaterials

Depending upon the severity of the *environmental stresses*, geomaterials might experience *moisture loss, movement and/or redistribution* within. These mechanisms are responsible for geomaterials to acquire *unsaturated-/partially-/variably-saturated* state. This state of geomaterials is, primarily, dictated by the *suction* present in them. The variation of suction with the water content is commonly known as the soil-water characteristic/retention curve, SWCC/SWRC. A tensiometer, as depicted in Fig. 34, can be employed for soil suction measurement (Singh and Kuriyan, 2003; Sreedeeep and Singh, 2005a,b). However, to overcome the limitations associated with the tensiometer measurements, which are limited to  $\approx 100$  kPa, Thakur and Singh (2005) and Thakur et al. (2005,2006) have developed a methodology, which facilitates measurement of suction up to a few Mega-Pascal. Subsequently, several methodologies were developed for measuring suction of the soil mass (Hanumantha and Singh, 2010b; Sreedeeep and Singh, 2011; Jayanth et al., 2013; Kannan et al., 2013; Iyer et al., 2018a,b; Patent No. 297124).



**Fig. 34. Details of the instrumentation for measuring soil suction by employing a tensiometer**

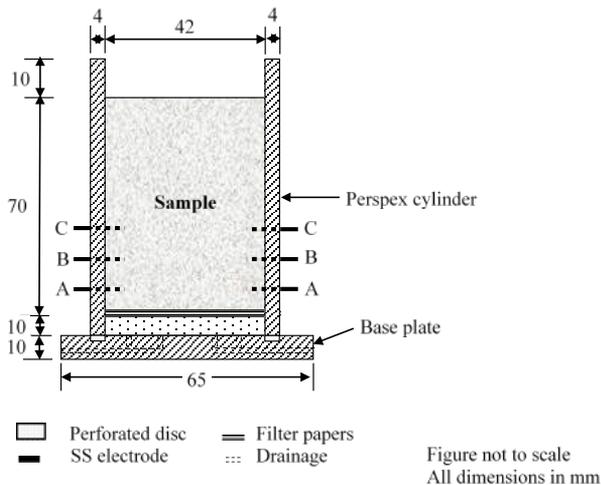
Unsaturated properties of the soil mass such as water retention and hydraulic conductivity also control the contaminant migration in it (Singh et al., 2001b; Singh and Kuriyan, 2002,2003). Earlier studies reveal that these characteristics mainly depend on the mineralogy, water content, dry-density and degree of saturation of the soil (Rao et al., 2011, Patent No. 277842). With this in view, various in-situ, laboratory techniques, empirical relationships and Pedo-transfer functions (PTFs), have been developed by the researchers to establish soil-water characteristic curve, SWCC, (Thakur et al., 2005,2006,2007; Shah et al., 2006; Sreedeeep and Singh, 2006b; Jayanth et al, 2012). However, these techniques are tedious, time-consuming, soil specific, require expensive experimental setups and resort to destructive and invasive techniques for determining soil moisture content (Sreedeeep and Singh, 2005b). Also, most of these techniques suffer particularly in regeneration of the in-situ soil conditions and boundary conditions, in an exceptionally small models (Borana et al., 2018). Usually, the SWCC is employed for estimating hydraulic conductivity of the unsaturated soil mass. Hence, techniques that would be helpful in (a) creating the unsaturated state of the soil mass and (b) determination of the unsaturated hydraulic conductivity of the soils, under laboratory and field conditions directly, were developed, as described below.

7.1 Laboratory Simulation of Unsaturated State of Geomaterials

The first step in this direction would be to create unsaturated state of the soil mass by (i) employing a geotechnical centrifuge, (ii) imposing thermal flux and (iii) employing a pressure membrane extractor (PME), and determining its instantaneous moisture content by conducting electrical measurements (i.e., voltage across two points in the soil mass). The utility of these techniques for establishing the SWCC, easily and that too in a nondestructive and noninvasive manner has been demonstrated (Hanumantha Rao and Singh, 2010a,b; Rao and Singh, 2012).

7.1.1 By Centrifugation

The test setup depicted in Fig. 35 was fabricated to attain an unsaturated state of the sample, by centrifugation, followed by establishing the SWCC. As depicted in the figure, point electrodes were fitted diagonally opposite to each other and the voltage across each pair of electrodes,  $V$ , was measured by employing the soil conductivity meter developed by Rajeev and Singh (2004).

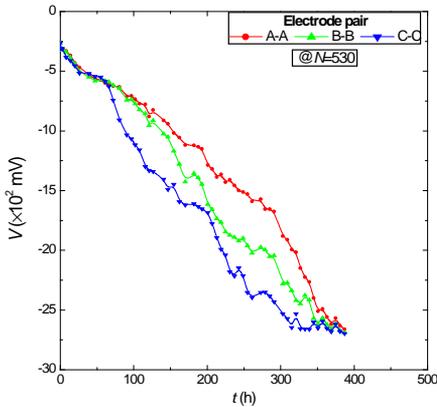


**Fig. 35. The setup used for creating unsaturated state of the soil mass by centrifugation**

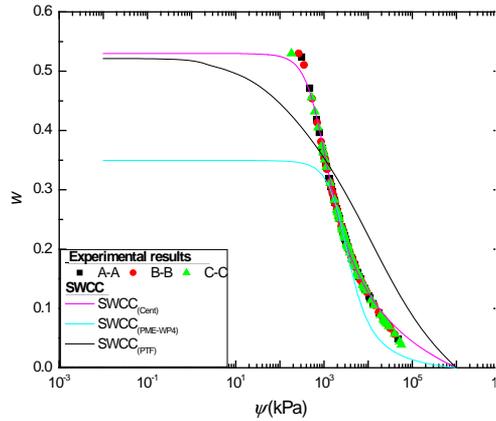
The soil sample (in a slurry form) was poured into the perspex cylinder and later centrifuged for different centrifugation efforts (time and  $g$ -level, designated as  $N$ ). Due to centrifugation, water drains out of the sample and it acquires the *unsaturated* state.

At the end of each centrifugation effort, the setup was taken out of the centrifuge and voltage,  $V$ , across different electrode pairs was measured. Later, a stainless steel ring was used for retrieving a specimen of the soil for measuring its suction,  $\psi$ , by employing a dewpoint potentiometer, WP4, (Shah et al., 2006). Subsequently, the ring containing the soil specimen was placed in an oven to determine its moisture content,  $w$ . Fig. 36(a) depicts the variation of the measured voltage,  $V$ , across electrode pairs A-A, B-B and C-C, with time,  $t$ . For the sake of brevity, results of  $N=530$ , only, are being presented. It can be observed from the trends depicted in the figure,  $V$  decreases as  $t$  increases, which indicates that the resistance of the soil mass increases, due to the expulsion of water as a result of enhanced centrifugation efforts. This observation further indicates that soil mass achieves certain unsaturated state after each centrifugation (Poulose et al., 2000; Singh et al., 2001; Singh and Kuriyan, 2002).

It was also observed that the difference between the values of  $V$  corresponding to a given time between two electrode pairs decreases for higher values of  $N$ . This indicates that centrifugation at higher  $N$  creates a uniform state of moisture in the sample in a short duration of time. The results have been employed for developing the soil-water characteristic curve,  $SWCC_{(Cent)}$ , using the Fredlund and Xing (1994) best fit. Further, to demonstrate the utility of the proposed technique, the obtained  $SWCC_{(Cent)}$  was compared vis-à-vis the  $SWCCs$  reported by researchers (Shah et al., 2006),  $SWCC_{(PME-WP4)}$ , and obtained from the pedo-transfer function, PTF, proposed by Fredlund et al., (1997),  $SWCC_{(PTF)}$ , as depicted in Fig. 36(b). A reasonably good match between the  $SWCC_{(Cent)}$  and  $SWCC_{(PME-WP4)}$  can be noticed.



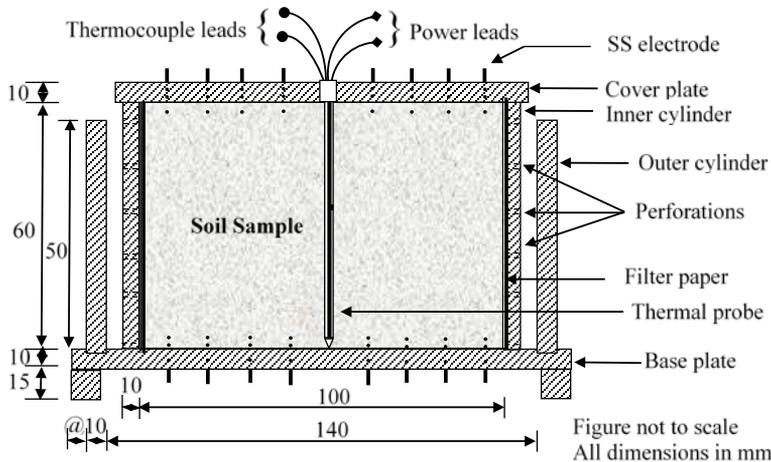
**Fig. 36(a).** Variation of the measured voltage across different electrodes with time



**Fig. 36(b).** Validation of the methodology for obtaining SWCC by centrifugation of the sample

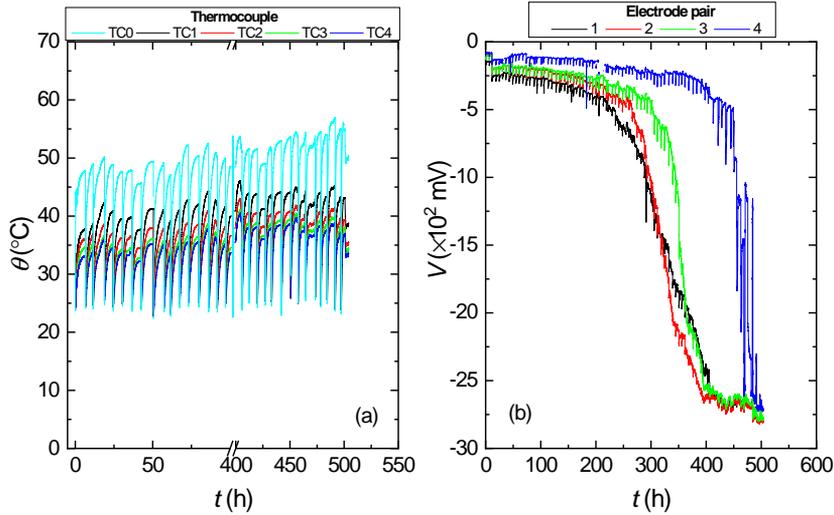
### 7.1.2 By Imposition of Thermal Flux

The test setup depicted in Fig. 37 was fabricated, as an alternative to the setup depicted in Fig. 35, to create unsaturated samples and for establishing the SWCC. The *laboratory thermal probe* is the most important part of this setup.



**Fig. 37.** The setup used for creating unsaturated state of the soil mass by thermal flux imposition

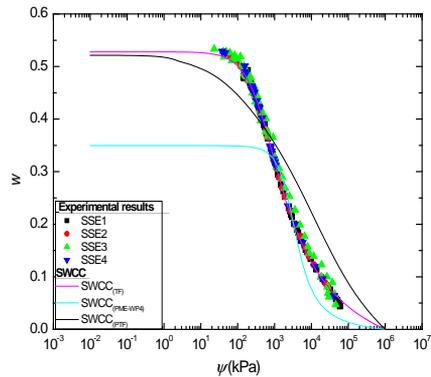
The curved surface of the inner cylinder is perforated in the radial direction. Due to imposition of the thermal flux, water from the soil sample permeates through these perforations and gets collected in the outer cylinder. The electrodes that are fitted on top and bottom plates facilitate measurement of the  $V$  with the help of a soil conductivity meter. Also, the cover plate has been provided with four 3 mm diameter holes for fitting thermocouples, which are placed at different radial distances from the probe. Fig. 38(a) depicts the variation of temperature,  $\theta$ , and measured voltage,  $V$ , with time,  $t$ , obtained by imposing a heat flux,  $Q$ , of 59.5 W/cm, through the thermal probe.



**Fig. 38. The variation of (a) temperature and (b) voltage across different electrodes, due to imposition of thermal flux**

It can be observed from Fig. 38(a) that  $\theta$  varies with time,  $t$ , and the maximum temperature built up in the soil mass is about 30 to 40 °C (except for the surface of the probe on which temperature rise is up to 50 to 60 °C). It can also be noticed that, as expected, the rise in  $\theta$  is least for the thermocouples located at higher radial distances, for a given  $t$ . Trends depicted in Fig. 38(b) indicate that in general,  $V$  decreases with an increase in  $t$ . This phenomenon can be attributed to the increase in resistance of the soil mass, due to thermal flux induced moisture migration and hence creation of its unsaturated state. Subsequently,  $w$  and  $\psi$  for different values of  $V$  were computed, and the results are presented in Fig. 39. This data can be employed for developing the soil-water characteristic curve due to thermal flux,  $SWCC_{(TF)}$ , using the Fredlund and Xing

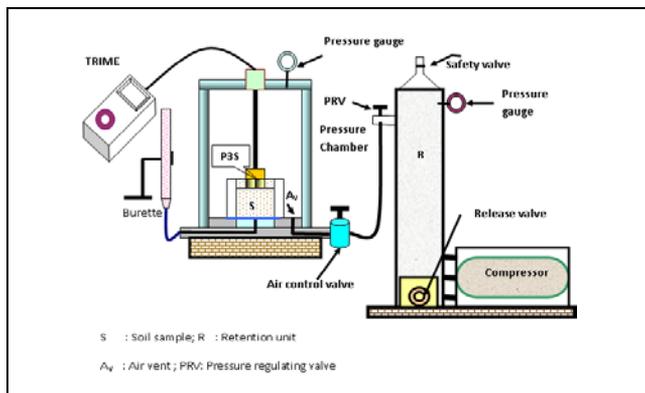
(1994) best fit. Further, to demonstrate utility of the proposed technique the  $SWCC_{(TF)}$  was compared vis-à-vis the  $SWCC$ s reported by researchers (Shah et al., 2006),  $SWCC_{(PME-WP4)}$ , and obtained from the pedo-transfer function proposed by Fredlund et al., (1997),  $SWCC_{(PTF)}$ , as depicted in the Fig. 39. A reasonably good match between the  $SWCC_{(TF)}$  and  $SWCC_{(PME-WP4)}$  can be noticed from this figure.



**Fig. 39. Validation of the methodology for obtaining SWCC by imposition of thermal flux on sample**

### 7.1.3 By Employing Pressure Membrane Extractor (PME)

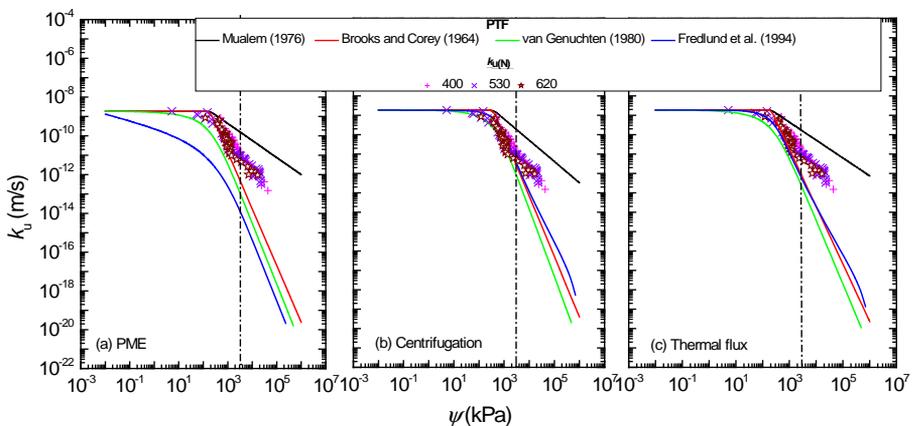
The PME unit, employed for creating different moisture states of the specimen, works in the pressure range of 0-1500 kPa and consists of a stainless steel pressure chamber and a base plate, refer Fig. 40.



**Fig. 40. The setup used for creating unsaturated state of the soil mass by employing a pressure membrane extractor**

Four PVC rings (60 mm internal diameter and 20 mm height) were used as specimen holders. These rings contain two pairs (1-1 and 2-2) of stainless steel point electrodes of 2 mm diameter, located diagonally opposite to each other. This arrangement helps in measuring voltage across two points,  $V$ , in the soil specimen. Subsequently, as explained earlier, its moisture content,  $W$  was also determined. However, this technique yields the response of the soil mass only up to the applied pressure, which is equal to suction,  $\psi \leq 1500$  kPa. Hence, the utility and efficiency of these three techniques for establishing the SWCC has been demonstrated. The most important feature of these techniques is that just by measuring the voltage across two points in the soil mass, its SWCC can be established, quite easily, and that too in a nondestructive and noninvasive manner. It is recommended that extensive experimentations should be conducted to generalize the suitability of the proposed technique for establishing SWCC of different types of soils.

Further, as depicted in Fig. 41, the unsaturated soil hydraulic conductivity,  $k_u$ , obtained by employing these three techniques and the PTFs, available in the SoilVision database<sup>®</sup>, were compared. It must be appreciated that  $k_{u(N)}$  are directly measured values, by adopting the methodology proposed by Singh and Kuriyan (2003), and not mathematically predicted. It's interesting to note that for all practical purposes,  $k_{u(N)}$  of the soil mass does not depend on  $N$ .

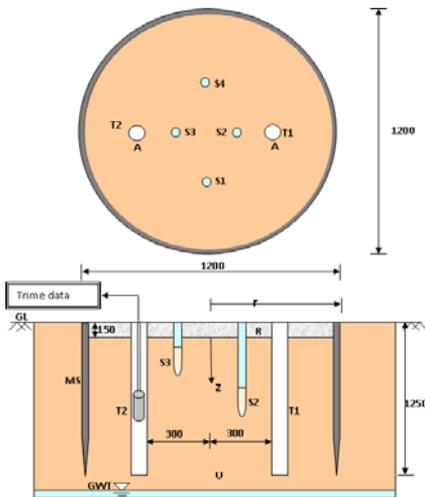


**Fig. 41. Evaluation of different techniques employed for estimating unsaturated soil hydraulic conductivity**

Hence, it can be said that these methodologies are very useful and handy for establishing the soil-water characteristic curve and for estimating the hydraulic conductivity of unsaturated soils, either directly (in case of centrifugation) or indirectly (i.e. by employing pedo-transfer functions, PTFs).

### 7.2. In-situ Lysimeter

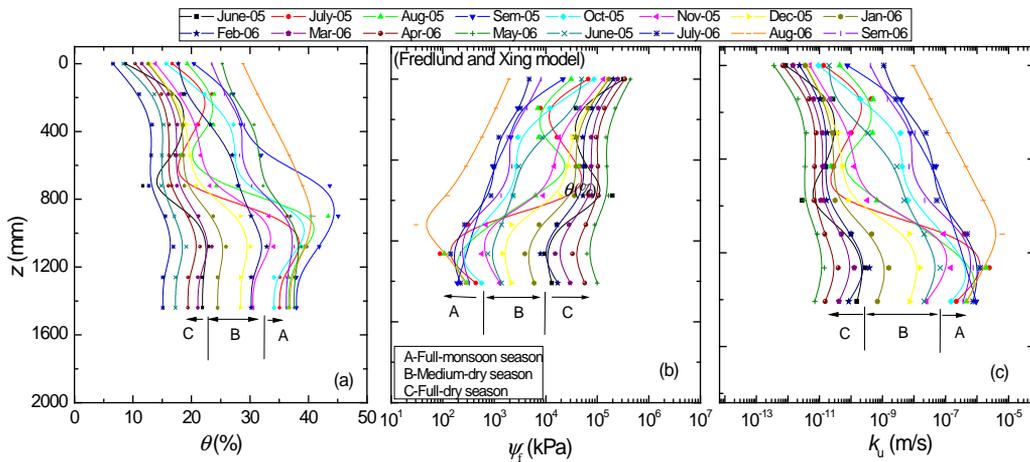
Laboratory column studies (under normal and accelerated gravity environments), numerical modeling and pedo-transfer functions (Thakur et al., 2007) have been employed by the researchers to determine hydraulic conductivity of unsaturated soils (Singh and Kuriyan, 2003; Hanumantha Rao et al., 2009) and studying contaminant transport (Rao et al., 2011) in them. However, these methods suffer from several limitations, particularly, as far as regeneration of in-situ soil conditions and boundary conditions, in an exceptionally small model, are concerned. Under these circumstances, an in-situ lysimeter (Fig. 42, legends used in the figure are explained in the table) has been found to be quite useful for conducting the investigations to obtain hydraulic conductivity of fully- or partially-saturated soils (Hanumantha Rao et al., 2009; Rao et al., 2011). A tubular TDR probe, has been employed for determining the volumetric moisture content,  $\theta$ , of the soil as a function of depth,  $z$ , by inserting it into access tubes T1 and T2.



Symbol	Description	Purpose
A	Access tube	Moisture detection probe
GWT		
R	Spiked soil	To simulate contaminated soil with Tritium
S1	Suction sampler (r,z)	(30, 40)
S2		(10, 30)
S3		(10, 20)
S4		(30, 10)
MS	Mild steel ring	For creating a control volume of approximately 1 m <sup>3</sup>
T1 & T2	Access tubes for TDR probe	For determining in-situ moisture content of the control volume
U	Undisturbed soil	Control volume of the soil

Fig. 42. Details of the in-situ lysimeter

Fig. 43 depicts spatial variation in the volumetric moisture content,  $\theta$ , of the soil mass corresponding to different monsoon seasons (full-monsoon, medium-dry and dry seasons). Trends depicted in the Fig. 43 (a) show that  $\theta$  varies over a wide range from 5 %, during the fully-dry season to 45%, during the full-monsoon season, which correspond to retention and saturation capacity of the soil mass, respectively. It can also be noticed from the Fig. 43 (a) that  $\theta$  increases with depth  $z$ . However, a significant scatter in  $\theta$  values can be observed for  $z < 300$  mm, only. Further, since unsaturated soil hydraulic conductivity,  $k_u$ , is a function of suction,  $\psi$ , efforts were made to determine suction of the soil mass in the lysimeter  $\psi_f$  by employing SWCC, with the help of Fredlund and Xing (1994) fitting function, to the corresponding measured  $\theta$  values, as depicted in Fig. 43(b). Hence, by knowing  $\psi_f$ ,  $k_u$  was computed by using the PTF proposed by Fredlund et al. (1994), as depicted in Fig. 43(c).



**Fig. 43. The variation of (a) volumetric moisture content, (b) suction, and (c)  $k_u$ , with  $z$  for different monsoon seasons**

## 8. Role of Environmental Stresses: Cracking of Fine-grained Soils

The cracking of fine-grained soils, can be attributed to the loss of moisture, as a result of its exposure to severe *environmental stresses*, as discussed above. These conditions might cause a non-uniform distribution of moisture and temperature, which might result in an imbalance of the internal energy. The loss of moisture from the fine-

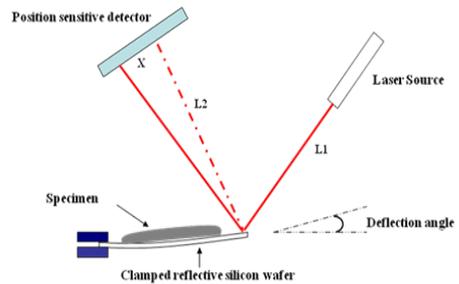
grained soils results in their shrinking and subsequently cracking, if they are restrained from undergoing volumetric changes, termed as shrinkage. It has been observed that *cracking* of the fine-grained soils influences its overall engineering behavior (in particular the hydraulic conductivity, consolidation, compressibility and shear strength characteristics). Hence, it becomes mandatory to investigate the mechanism(s) of soil cracking and the parameters that influence it significantly.

In this context, several experimental techniques viz., triaxial tests, direct tensile tests or suction measurements, have been developed and employed by earlier researchers to measure the tensile strength of fine-grained soils (Venkataramana et al., 2009) and to study their cracking characteristics (Uday and Singh, 2013a,b; Patent No. 277862). However, these studies yield results that are soil specific and dependent on the methodology adopted. Apart from this, due to the bulk form of the sample, the sample heterogeneity (i.e., both in terms of the density and the moisture content) influences test results to a great extent. Under this situation, and in the absence of guidelines regarding sample thickness, determination of the tensile strength of expansive clays by employing their thin samples (1 mm to 5 mm) appears to be an excellent alternative. Efforts have been made to determine *tensile strength* of fine-grained soils, by resorting to techniques that deal with measurement of the (a) deflection undergone by a silicon wafer, (refer Fig. 44a and b), due to air-drying of a thin film of these clays, with the help of a laser beam (Shinde et al., 2012), and (b) measuring suction by employing a dewpoint potentiometer, WP4 (Thakur et al., 2006). Results obtained from these techniques were critically evaluated vis-à-vis those obtained from triaxial tests and empirical relationships available in the literature. It has been observed that there is a unique relationship between the results of the thin and thick samples and hence thin samples of expansive clays can be employed for determining their tensile strength (Tej and Singh, 2013; Uday et al., 2014,2015). This shows the usefulness of the proposed techniques for measuring the *tensile strength* of the fine-grained, that too expansive soils (Bharat et al., 2013; Jha and Sivapullaiah, 2017).

The photograph and the schematic of the experimental set up used to measure the tensile stress developed in the thin film are depicted in Fig. 44(a) and (b), respectively. As depicted in Fig. 44(b), L1 and L2 are incident and reflected rays, respectively, and X is the displacement of the laser beam on the detector due to certain deflection of the wafer.

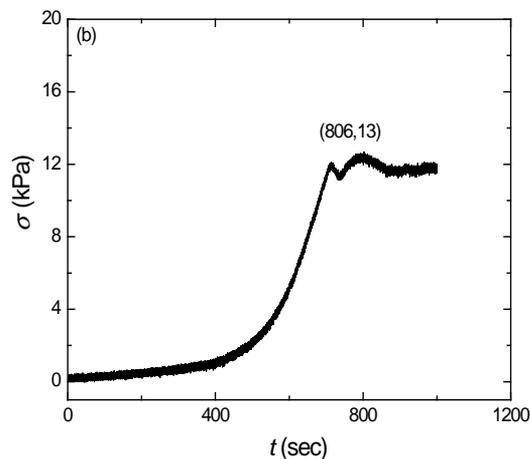


**Fig. 44(a). The experimental setup**



**Fig. 44(b). The basic concept of the tensile strength determination**

Subsequently,  $X$ , and stress,  $\sigma$ , as a function of time are recorded for thin film, as depicted in Fig. 45. The stress corresponding to the peak value yields the tensile strength of the specimen.



**Fig. 45. Experimental results depicting the variation of stress with time for a thin clay specimen**

This methodology of determining tensile strength of fine-grained soils, in tandem with application of dewpoint potentiometer, WP4, for expansive soils, in particular, corresponding to different liquid to solid ratios has been found to be quite useful. Results obtained from these methodologies were critically evaluated vis-à-vis those obtained from the triaxial tests and the empirical relationships reported in the literature. It has been observed that there is a unique relationship between the results

of the thin and thick samples. Hence, the proposed methodology can be employed for determining *tensile strength* of the fine-grained soils, just by knowing their mineralogical composition and the water content. However, extensive studies should be conducted on different minerals, and soils of different mineralogical composition to generalize these findings.

## 9. Concluding remarks & the Path Ahead

Various instrumentation techniques developed by *THE ENVGEOs* for providing solutions to various geoenvironmental issues have been presented in this paper. These techniques facilitate obtaining appropriate solutions to storage and safe disposal of hazardous and toxic wastes in geomaterials and thus minimizing the adverse effect on the geoenvironment. Though most of these techniques are found to be functioning quite satisfactorily in the laboratory environment, extensive in-situ investigations must be carried out that too on different types of geomaterials to demonstrate their robustness and suitability.

Though the (limited) contents of this paper must have successfully highlighted the necessity of *interdisciplinary research* in the field of contemporary geotechnical engineering, the following needs are to be addressed:

1. The impact of *biogeochemical (bio-mediated)* processes viz., *decay, degradation* and *upgradation*, that occur in geoenvironment, on *geomaterials*, should be studied and included in the analysis of geotechnical problems. The modification of engineering characteristics of the coarse-grained soils due to *biologically induced mineralization* is an excellent example, in this context.
2. The role of *geomaterial and pore-fluid* interaction in the soil matrix at the micro-structure level to address the *real-life* situations such as offshore landslides, sedimentation, migration of the fines and slurry transportation for various industrial applications viz., dredging and mineral processing, should be studied. In order to address this, investigations are warranted on the rheological behavior of soil-water systems corresponding to *water content* higher than the *liquid limit* (Shetty et al., 2018).

3. Accelerated decomposition of (a) unsegregated municipal solid waste, MSW, in a bioreactor landfill and (b) organic content rich soils, should be studied extensively, in line with the philosophy, *waste to energy*. Also, pilot-scale studies to incorporate the effect of *heterogeneity* on various *physico-biochemical* properties associated with these geomaterials, with the help of *sensors* and *probes*, should also be attempted. Apart from the laboratory and field investigations, numerical modeling related to *coupled processes* that are responsible for decomposition of these geomaterials must be attempted (Reddy et al., 2017b). Attempts should also be made to utilize the non-degradable fraction of the residues, which is a source of refuse-derived fuel, RDF, from the MSW.
4. The MSW landfill mining, a precursor to their *Rehabilitation and reclamation of the derelict land*, is shaping up as a giant ‘techno-socio-economical’ issue, which should be handled with great diligence. However, it should be realized that the *landfill mining* would result in extreme volumes of the *landfill mined residues* and subsequently their handling, segregation, storage, transportation and utilization for development of *infrastructure* and *soil conditioning* needs to be addressed scientifically. Appropriate *standards, guidelines and policies* need to be developed in this context.
5. The industrial by-products, IBPs, viz., bauxite residues, slags, fly ash, mine overburden dumps, tailings, plastics, etc., should be treated as the *man-made resource* for *sustainable development* (Mohamed, 2015; Reddy et al., 2018). This could be attained by developing proper technology(ies) based on extensive R&D for *bulk utilization* of the IBPs. Attempts should also be made towards the *creation and implementation* of appropriate policies. In this context, the *hazard* and *toxicity* zonation of the entire country would be advantageous.
6. Geomechanical modeling of *multi-phase* geomaterials viz., permafrost, gas hydrates bearing sediments, thermally stabilized soils, the municipal solid waste, etc., which is still in its nascent stage, should be taken up.

However, in order to accomplish such contemporary research, indigenous development of the test setups and paraphernalia, which also falls in line with the Government of India initiative, MAKE IN INDIA, is the dire need of the hour. Furthermore, most of these issues could easily be taken up as *startups* by the

professionals, who are concerned about the *generations to follow*, and for whom the sky is the limit. This would also remind them that the path ahead is not so easy to tread but *if they desire to blossom like a rose in the garden, they have to learn the art of adjusting with the thorns.*

I would also like to add here that audiences' critical comments and suggestions are utmost welcome without which it would be difficult for *The ENVGEOs* to march ahead and follow the philosophy *Never stop learning, because life never stops Teaching* .... Furthermore, the outcome of these initiatives should be *adopted, challenged and improvised* in view of the fact that *Life is a Camera, Focus on what's important Capture the good times, Develop from the negatives and if things don't work out, Take another Shot...*

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